

Chapter 2

Atoms and Molecules: The Chemical Basis of Life

- The chemical similarities among all organisms on Earth provide strong evidence for the evolution of all organisms from a common ancestor
- Molecular biology: the study of chemistry and physics of the molecules that constitute living things
- Inorganic compounds: small, simple substances
 - Biologically important groups include water, simple acids and bases, and simple salts
- Organic compounds: generally large, complex carbon- containing compounds

2.1 Elements and Atoms

- Elements: substances that can't be broken down into simpler substances by ordinary chemical reactions
 - Each element has a chemical symbol
 - Oxygen, Carbon, Hydrogen, and Nitrogen are the four primary elements that make up >96% of the mass of most organisms
- Atom: the smallest unit of an element that retains that element's chemical properties
 - Made up of tiny subatomic particles: neutrons, protons and electrons
 - Matter: anything that has mass and takes up space

An Atom is Identified by its Number of Protons

- Atomic number: the fixed number of protons in the atomic nucleus (# of protons)
 - Determines an atom's identity and defines the element
- Periodic table: a chart of the elements
 - Arranged in order by atomic number and chemical behavior



- Bohr models represent the electron configurations of elements as a series of concentric rings
- In a neutral atom, protons will equal electrons

The Chemical Basis Of Life Chapter 3 Packet Answers

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Velavancorresponding**



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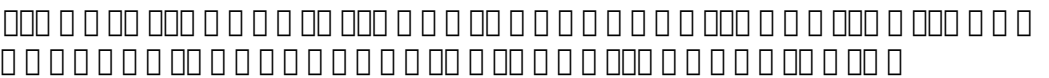
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