

## REVIEW OF BASIC CHEMISTRY ANSWER KEY

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1. Name the following elements. Spelling counts:

H	<i>hydrogen</i>
S	<i>sulphur</i>
Cl	<i>chlorine</i>
C	<i>carbon</i>
N	<i>nitrogen</i>
Na	<i>sodium</i>
Pb	<i>lead</i>
Ag	<i>silver</i>

2. Write the symbols for the following elements.

sodium	<i>Na</i>
phosphorus	<i>P</i>
fluorine	<i>F</i>
magnesium	<i>Mg</i>
potassium	<i>K</i>
calcium	<i>Ca</i>
zinc	<i>Zn</i>
iron	<i>Fe</i>

3. Using a periodic table, record the atomic number for the following elements:

gold (Au)    **79**                      copper (Cu)    **29**

4. Using a periodic table, record the atomic mass for the following elements to one decimal place:

chlorine (Cl)    **35.5**                      calcium (Ca)    **40.1**

5. Complete the following table. Use the information provided in the chart – not a periodic table – to determine atomic masses, but you may refer to a periodic table to name the element.

Element	Atomic Number	Atomic Mass	Protons	Neutrons	Electrons
<b>Lead</b>	<b>82</b>	<b>207</b>	<b>82</b>	125	<b>82</b>
<b>Argon</b>	<b>18</b>	<b>56</b>	18	40	<b>18</b>
Barium	<b>56</b>	137	<b>56</b>	<b>61</b>	<b>56</b>
<b>Tin</b>	<b>50</b>	112	<b>50</b>	<b>62</b>	50

6. Match the unit with its description:

<u><b>F</b></u>	1. amount of substance
<u><b>C</b></u>	2. density
<u><b>B</b></u>	3. mass
<u><b>D</b></u>	4. molar mass
<u><b>A</b></u>	5. energy
<u><b>E</b></u>	6. volume

A.	joule
B.	g
C.	g/mL
D.	g/mole
E.	L
F.	mole

# World Of Chemistry Basic Review Sheet

**Tobias Bleicker**



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