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web jul 23 2023 the covalent radius r cov is a measure of the size of an atom that forms part of one covalent bond it is usually measured either in picometres pm orangetroms \mathring{A} with 1 \mathring{A} 100 pm in

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web jun 14 2022 explain your answer what are the predominant interactions when oppositely charged ions are far apart at internuclear distances close to r0 very close together at a distance that is less than the sum of the ionic radii several factors contribute to the stability of ionic compounds