compound (II). Compound (II) on disproportionation in neutral and acidic solution gives a purple coloured compound (C). Identify A, III and C and write the reaction involved when compound (C) is heated to 51.1 K.

Ober.

(i) Eⁿ Mⁿ / Mⁿ values for the first series of transitionalements are given below. Asswer the question that follows:

E" (V)	177	W	Circ	Man	Fer	THE STREET
34 7 34	40.33	40.26	-0.45	+1.57	+0.37	-1.93

Educately the two strongest entidizing agents in the aqueous autotion from the above date.

- iii) Copper (I) loss is not known in aqueous solution.
- in) The highest enableton state of a metal is exhibited in its evolute.

OW.

- (a) Write balanced equations to appresent what happens when
 - (ii) Cu is treated with KL.
 - (3) Acidified perassium distinguise solution is reacted with imm (II) solution. (lomic equation)

1200

 The figure green before illustrates the first ionization enthalpies of first, second and third spring of transition elements. Answer the question that follows:



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