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Valence Electrons

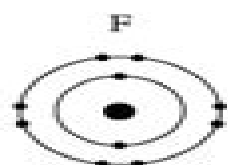
1 How many electrons are present in the following elements?

- a Carbon ____ b Chlorine ____ c Indium ____ d Calcium ____
e Tungsten ____ f Antimony ____ g Titanium ____ h Tin ____

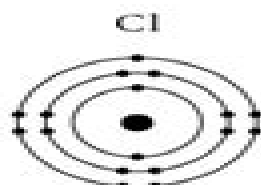
2 Using the periodic table, determine the number of valance electrons for the elements listed below.

- a Copper ____ b Xenon ____ c Hydrogen ____ d Aluminum ____
e Barium ____ f Iodine ____ g Carbon ____ h Phosphorous ____

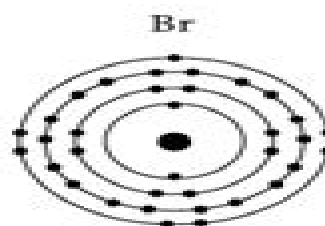
3 Circle the valence electrons only. What is the atomic number?



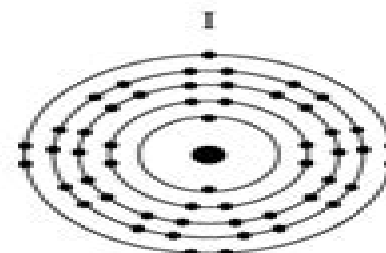
Atomic number(Z) =



Atomic number(Z) =



Atomic number(Z) =



Atomic number(Z) =

4 Fill in the table with correct values.

Element	Ion symbol	Number of Protons	Number of electrons	Charge	Ion type
Fluorine	F ⁻			-1	Anion
		53	54		
		16		-2	
Potassium				+1	
Calcium					

Valence Electrons And Ions Worksheet Answer Key

Sir Tarun Rupani



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web of nuclear science and engineering is a key reference for any physicists or engineer radiochemistry or nuclear chemistry is the study of radiation from an atomic or molecular perspective including elemental transformation and reaction effects as well as physical health and medical properties

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web sodium 23 in a sample of NaCl is subjected to neutron bombardment in a nuclear reactor to produce ^{24}Na when removed from the reactor the sample is radioactive with beta activity of $2.54 \times 10^4 \text{ dpm}$ the decrease in radioactivity over time was studied producing the following data