

Student Exploration: Calorimetry Lab

Vocabulary: calorie, calorimeter, joule, specific heat capacity

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

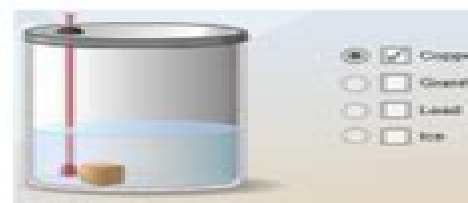
1. The Latin word *calor* means “heat,” and *meter* comes from the Greek word meaning “to measure.” What do you think a **calorimeter** does? A device used to measure the quantity of heat transferred to or from an object.

2. Where have you heard the word **calorie** before? What do you think a calorie is? A unit of measurement/energy.

Gizmo Warm-up

A calorimeter is an insulated container filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up.

Calorimeters can be used to find a substance's **specific heat capacity**. You will use the *Calorimetry Lab* Gizmo to determine the specific heat capacities of various substances.



1. On the SIMULATION pane, select **Copper**. Use the slider to set its **Mass** to 200 g. Set the **Water mass** to 200 g. Check that the **Water temp** is set to 30.0 °C and the copper's **Temp** is 90 °C. Select the GRAPH tab, and click **Play** ().

A. What was the Final temperature of the copper and the water? 34.96

B. How much did the temperature of the copper change? 55.04

C. How much did the temperature of the water change? 4.96

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E Durkheim



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