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Date: _____

Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? Share

2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

Share 2

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** () to see the electrons orbiting the nucleus of each atom.



1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have? 7

2. Click **Pause** (). Drag an electron from the left atom to the right atom. Click **Play**.

What happens? it orbits both atoms

3. Click **Pause**, drag an electron from the right atom to the left, and then click **Play**.

What happens now? it orbits both atoms

Student Exploration Covalent Bonds Answer For Packet

N Noddings



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Table of Contents Student Exploration Covalent Bonds Answer For Packet

1. Understanding the eBook Student Exploration Covalent Bonds Answer For Packet
 - The Rise of Digital Reading Student Exploration Covalent Bonds Answer For Packet
 - Advantages of eBooks Over Traditional Books
2. Identifying Student Exploration Covalent Bonds Answer For Packet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Student Exploration Covalent Bonds Answer For Packet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Student Exploration Covalent Bonds Answer For Packet
 - Personalized Recommendations
 - Student Exploration Covalent Bonds Answer For Packet User Reviews and Ratings

- Student Exploration Covalent Bonds Answer For Packet and Bestseller Lists
- 5. Accessing Student Exploration Covalent Bonds Answer For Packet Free and Paid eBooks
 - Student Exploration Covalent Bonds Answer For Packet Public Domain eBooks
 - Student Exploration Covalent Bonds Answer For Packet eBook Subscription Services
 - Student Exploration Covalent Bonds Answer For Packet Budget-Friendly Options
- 6. Navigating Student Exploration Covalent Bonds Answer For Packet eBook Formats
 - ePub, PDF, MOBI, and More
 - Student Exploration Covalent Bonds Answer For Packet Compatibility with Devices
 - Student Exploration Covalent Bonds Answer For Packet Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Student Exploration Covalent Bonds Answer For Packet
 - Highlighting and Note-Taking Student Exploration Covalent Bonds Answer For Packet
 - Interactive Elements Student Exploration Covalent Bonds Answer For Packet
- 8. Staying Engaged with Student Exploration Covalent Bonds Answer For Packet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Student Exploration Covalent Bonds Answer For Packet
- 9. Balancing eBooks and Physical Books Student Exploration Covalent Bonds Answer For Packet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Student Exploration Covalent Bonds Answer For Packet
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Student Exploration Covalent Bonds Answer For Packet
 - Setting Reading Goals Student Exploration Covalent Bonds Answer For Packet
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Student Exploration Covalent Bonds Answer For Packet
 - Fact-Checking eBook Content of Student Exploration Covalent Bonds Answer For Packet
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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