

Straighterline Chemistry Final- deck 3

Which is more soluble in an acidic solution than in pure water?

Select one:

- a. CuI
- b. PbCl_2
- c. $\text{Ca}_3(\text{PO}_4)_2$
- d. NaNO_3
- e. NaBr - ANSc. $\text{Ca}_3(\text{PO}_4)_2$

What is the name of the principle of selective precipitation used to identify the types of ions present in a solution?

Select one:

- a. Ionization
- b. Selective ion precipitation
- c. Selective ion typing
- d. Limited precipitation
- e. Qualitative analysis - ANSe. Qualitative analysis

A 10.0-mL sample of 0.75 M $\text{CH}_3\text{CH}_2\text{COOH}$ is titrated with 0.30 M NaOH . What is the pH of the solution after 22.0 mL of NaOH have been added to the acid?

$[\text{Ka}(\text{CH}_3\text{CH}_2\text{COOH}) = 1.3 \times 10^{-5}]$

Select one:

- a. 5.75
- b. 4.94
- c. 4.83
- d. 4.02
- e. 3.95 - ANSa. 5.75

Methyl red is a common acid-base indicator. It has a Ka equal to 6.3×10^{-6} . Its un-ionized form is red and its anionic form is yellow. What color would a methyl red solution have at $\text{pH} = 7.8$?

Select one:

- a. green
- b. red
- c. blue
- d. yellow
- e. violet - ANSd. yellow

A 25.0-mL sample of 0.10 M $\text{C}_2\text{H}_5\text{NH}_2$ (ethylamine) is titrated with 0.15 M HCl . What is the pH of the solution after 9.00 mL of acid have been added to the amine?

$[\text{Kb}(\text{C}_2\text{H}_5\text{NH}_2) = 6.5 \times 10^{-4}]$

Select one:

Straighterline Test Answer

Deborah L. Bandalos

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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Instructor's Manual with Test Bank [for] Basic Concepts of Chemistry, Fourth Edition Alan Sherman, 1988 *Teaching in the Study of Religion and Beyond* Russell T. McCutcheon, 2024-02-08 Drawing on their wide experience in the undergraduate classroom the contributors address basic but current issues in university teaching This book provides practical commentary and invites instructors to consider how to address the learning needs of their students while taking into account the wider structural requirements of administrations governments or credentialing agencies Consisting of about forty readable short entries on topics ranging from curriculum grading group work digital humanities and large lectures to learning management systems office hours online remote courses recruiting and seminars this book provides a wealth of practical help and reassurance to teachers working with undergraduate students This book is a valuable tool for early instructors in universities and colleges showing them how to impact a class's success It provides a critical background on the issues involved whilst also offering suggestions on how to navigate the competing demands on teachers *Measurement Theory and Applications for the Social Sciences* Deborah L. Bandalos, 2018-01-31 Which types of validity evidence should be considered when determining whether a scale is appropriate for a given measurement situation What about reliability evidence Using clear explanations illustrated by examples from across the social and behavioral sciences this engaging text prepares students to make effective decisions about the selection administration scoring interpretation and development of measurement instruments Coverage includes the essential measurement topics of scale development item writing and analysis and reliability and validity as well as more advanced topics such as exploratory and confirmatory factor analysis item response theory diagnostic classification models test bias and fairness standard setting and equating End of chapter exercises with answers emphasize both computations and conceptual understanding to encourage readers to think critically about the material **Jacaranda Science Quest 8 Australian Curriculum, 4e learnON and Print** Graeme Lofts, Merrin J. Evergreen, 2023-11-20 For junior secondary school age **Jacaranda Science Quest 8 for the Victorian Curriculum, LearnON and Print** Graeme Lofts, Merrin J. Evergreen, Sarah Beamish, 2020-11-11 Jacaranda Science Quest 8 Victorian Curriculum 2nd Edition learnON Print This combined print and digital title provides 100% coverage of the Victorian Curriculum for Science The textbook comes with a complimentary activation code for learnON the powerful digital learning platform making learning personalised and visible for both students and teachers The latest editions of Jacaranda Science Quest Victorian Curriculum series include these key features Content is completely revised and updated aligned to the Science 7-10 Victorian Curriculum and consistent across all platforms learnON eBookPLUS PDF iPad app and print Concepts are brought to life with engaging content diagrams and illustrations and digital resources including interactivities videos weblinks and projects Explicit development of Science Inquiry Skills SIS that is embedded throughout the content and questions Science Inquiry SkillBuilders and introduction to science SIS topics A brand new student eWorkbook and Practical

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