

Unit 4: Laboratory Techniques and their Application

Assignment C: Exploring manufacturing techniques and testing methods for an organic solid

P5 – Correctly prepare and test the purity of an organic solid and draw conclusions

The Preparation of Aspirin

Introduction – What is Aspirin?

Aspirin is a synthetic drug most commonly used in the United Kingdom, with the UK producing over 4 million tablets per year. The most common use of this tablet is to reduce fever symptom such as a cold or headache, it can also be prescribed to be used to reduce the chances of a heart attack. Aspirin is also commonly known as acetylsalicylic acid.

Formula of Aspirin – $C_9H_8O_4$

Aim

The aim of this experiment to produce aspirin a viable number of aspirin which can be filtered, and then I should be able to test the purity by testing its melting point, the preparation of the aspirin is the first step before testing the purity.

Equipment

Safety –

- Goggles – Protect eyes and face from possible chemical splashes.
- Lab coat – Buttoned up – to protect torso and clothes from contamination.
- Gloves – Reduces the exposure to the chemicals being used.

Experiment –

- 100g of 2-hydroxybenzenecarboxylic acid (OH- group – attached to the benzene ring)
- 100 ml of Ethanoic anhydride
- Concentrated sulphuric acid – this will act as the catalyst for this reaction.
- 100cm³ conical flask
- Spatula – Used to transfer the solid.

Unit Assignment Applications

**United States. Federal
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Intellectual Property Legal Opinions Thronson, Roth, Grossman, 1997

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