

Name: _____

Chemistry Practice: Balancing Chemical Equations

1. Balance the equations below. Remember this means only using coefficients. All formulas are written correctly.

- a.) $2 \text{NaHCO}_3 \longrightarrow \text{Na}_2\text{CO}_3 + \text{H}_2\text{O} + \text{CO}_2$
- b.) $\text{Zn} + \text{S} \longrightarrow \text{ZnS}$
- c.) $2 \text{H}_2 + \text{O}_2 \longrightarrow 2 \text{H}_2\text{O}$
- d.) $2 \text{FeO} + \text{C} \longrightarrow 2 \text{Fe} + \text{CO}_2$
- e.) $\text{Cl}_2 + 2 \text{KI} \longrightarrow 2 \text{KCl} + \text{I}_2$
- f.) $\text{B}_2\text{O}_3 + 3 \text{Mg} \longrightarrow 3 \text{MgO} + 2 \text{B}$
- g.) $\text{Ca} + 2 \text{H}_2\text{O} \longrightarrow \text{Ca(OH)}_2 + 2 \text{H}_2$
- h.) $4 \text{H}_2\text{O} + \text{Na}_2\text{B}_4\text{O}_7 + \text{H}_2\text{SO}_4 \longrightarrow 4 \text{H}_3\text{BO}_3 + \text{Na}_2\text{SO}_4$
- i.) $\text{CaC}_2 + 2 \text{H}_2\text{O} \longrightarrow \text{Ca(OH)}_2 + \text{C}_2\text{H}_2$
- j.) $\text{NH}_4\text{NO}_3 \longrightarrow \text{N}_2\text{O} + 2 \text{H}_2\text{O}$
- k.) $\text{H}_2\text{S} + \text{O}_2 \longrightarrow \text{SO}_2 + \text{H}_2\text{O}$
- l.) $\text{H}_2\text{SO}_4 + 2 \text{NaCl} \longrightarrow 2 \text{HCl} + \text{Na}_2\text{SO}_4$
- m.) $2 \text{NaCl} + 2 \text{H}_2\text{O} \longrightarrow 2 \text{NaOH} + \text{Cl}_2 + \text{H}_2$
- n.) $3 \text{BaO} + 2 \text{Al} \longrightarrow 3 \text{Ba} + \text{Al}_2\text{O}_3$
- o.) $3 \text{O}_2 \longrightarrow 2 \text{O}_3$
- p.) $\text{SiO}_2 + 2 \text{C} \longrightarrow 2 \text{CO} + \text{Si}$
- q.) $2 \text{PbS} + 3 \text{O}_2 \longrightarrow 2 \text{PbO} + 2 \text{SO}_2$
- r.) $2 \text{HgO} \longrightarrow 2 \text{Hg} + \text{O}_2$
- s.) $\text{C}_{12}\text{H}_{22}\text{O}_{11} + 12 \text{O}_2 \longrightarrow 12 \text{CO}_2 + 11 \text{H}_2\text{O}$
- t.) $\text{C}_3\text{H}_8 + 5 \text{O}_2 \longrightarrow 3 \text{CO}_2 + 4 \text{H}_2\text{O}$
- u.) $2 \text{FeCl}_3 + 3 (\text{NH}_4)_2\text{S} \longrightarrow 6 \text{NH}_4\text{Cl} + \text{Fe}_2\text{S}_3$
- v.) $\text{P}_2\text{O}_5 + 6 \text{NaOH} \longrightarrow 2 \text{Na}_3\text{PO}_4 + 3 \text{H}_2\text{O}$
- w.) $3 \text{BaCl}_2 + \text{Al}_2(\text{SO}_4)_3 \longrightarrow 3 \text{BaSO}_4 + 2 \text{AlCl}_3$
- x.) $4 \text{NH}_3 + 5 \text{O}_2 \longrightarrow 6 \text{H}_2\text{O} + 4 \text{NO}$
- y.) $2 \text{Na}_2\text{O}_2 + 2 \text{H}_2\text{O} \longrightarrow 4 \text{NaOH} + \text{O}_2$
- z.) $\text{Sn(ClO}_3)_4 \longrightarrow \text{SnCl}_4 + 6 \text{O}_2$

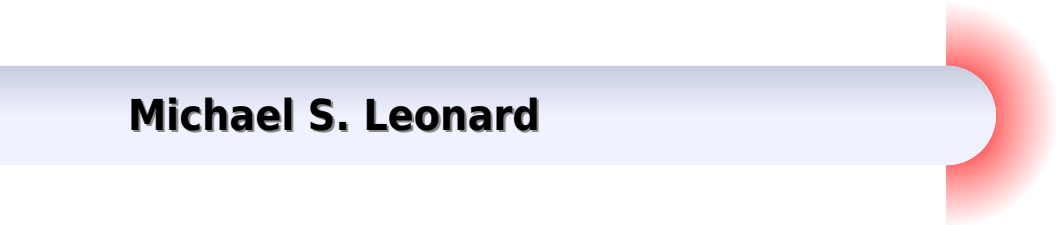
Text Problems to be Completed on the Other Side of this Worksheet:

*P. 173 #45,49,53,54,56,57,58,63, 65,66,67,71

*(Although the directions may say to write an "unbalanced" equation, write a balanced equation for each.)

Unit 6 Balancing Chemical Reactions 1 Key

Michael S. Leonard



Unit 6 Balancing Chemical Reactions 1 Key:

Basic Concepts of Chemistry Leo J. Malone, Theodore Dolter, 2008-12-03 Engineers who need to have a better understanding of chemistry will benefit from this accessible book. It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features. Each section focuses on the development and assessment of one or two specific objectives. Within each section a specific objective is included, an anticipatory set to orient the reader, content discussion from established authors and guided practice problems for relevant objectives. These features are followed by a set of independent practice problems. The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical related topics. Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance.

Physical Chemistry for Engineering and Applied Sciences Frank R. Foulkes, 2012-09-12 Physical Chemistry for Engineering and Applied Sciences is the product of over 30 years of teaching first year Physical Chemistry as part of the Faculty of Applied Science and Engineering at the University of Toronto. Designed to be as rigorous as compatible with a first year student's ability to understand the text, presents detailed step by step derivations of the equations that permit the student to follow the underlying logic and of equal importance to appreciate any simplifying assumptions made or mathematical tricks employed. In addition to the 600 exercises and end of chapter problems, the text is rich in worked non trivial examples many of which are designed to be inspiring and thought provoking. Step by step derivation of all equations enables the student to smoothly follow the derivation by sight and can be understood relatively easily by students with moderate skills and backgrounds in mathematics. Clear and accessible Physical Chemistry for Engineering and Applied Sciences includes The answers to all of the 112 worked examples, 99 exercises following many of the worked examples and 496 end of chapter problems. Topics not normally seen in introductory physical chemistry textbooks: ionic reaction rates, activities and activity coefficients or not regularly explained in much detail, electrochemistry, chemical kinetics with an eye on industrial applications. Special appendices that provide detailed explanations of basic integration and natural logarithms for students lacking a background in integral calculus. An in depth chapter on electrochemistry in which activities and activity coefficients are used extensively as required for accurate calculations.

Calculating Theoretical Yield Is Not Impossible!
Michael S. Leonard, 2025-02-25 Calculating theoretical and percent yield is a fundamental skill for the laboratory. This book primarily targets Organic Chemistry Laboratory courses at the high school or college and university level as a supplemental resource to help students master this skill. It begins with simple examples from everyday life, demonstrates the importance of balancing the equation, addresses the role of the mole in these computations, discusses different types of liquids, considers the role of significant figures and culminates with the planning of syntheses. There are suggestions for further reading as well as practice problems and questions to ensure mastery. Begins with examples from everyday life that enable students to

understand the concepts of theoretical and percent yield before applying those concepts to the laboratory Addresses the necessity of balancing the reaction equation the centrality of the mole in these calculations and the role of significant figures in reporting the answer Explains how to approach the calculations when using neat liquids or solutions The culmination of this text is the use of the same thought processes to plan the amounts of reactants needed for syntheses of desired quantities of product All of the problems in the book include detailed solutions with accompanying text to explain the answers and ancillaries also include suggestions for further reading Chemical Reactions and Chemical Reactors George W.

Roberts,2008-03-14 Focused on the undergraduate audience Chemical Reaction Engineering provides students with complete coverage of the fundamentals including in depth coverage of chemical kinetics By introducing heterogeneous catalysis early in the book the text gives students the knowledge they need to solve real chemistry and industrial problems An emphasis on problem solving and numerical techniques ensures students learn and practice the skills they will need later on whether for industry or graduate work **Material Balances for Chemical Reacting Systems** R.L. Cerro,B.G.

Higgins,S. Whitaker,2022-12-05 Written for use in the first course of a typical chemical engineering program Material Balances for Chemical Reacting Systems introduces and teaches students a rigorous approach to solving the types of macroscopic balance problems they will encounter as chemical engineers This first course is generally taken after students have completed their studies of calculus and vector analysis and these subjects are employed throughout this text Since courses on ordinary differential equations and linear algebra are often taken simultaneously with the first chemical engineering course these subjects are introduced as needed Teaches readers the fundamental concepts associated with macroscopic balance analysis of multicomponent reacting systems Offers a novel and scientifically correct approach to handling chemical reactions Includes an introductory approach to chemical kinetics Features many worked out problems beginning with those that can be solved by hand and ending with those that benefit from the use of computer software This textbook is aimed at undergraduate chemical engineering students but can be used as a reference for graduate students and professional chemical engineers as well as readers from environmental engineering and bioengineering The text features a solutions manual with detailed solutions for all problems as well as PowerPoint lecture slides available to adopting professors

HESI A2 Study Guide Tyler L. Jackson,2025-01-04 Are you preparing for the HESI A2 exam and feeling overwhelmed by the range of subjects you need to master Whether you re aiming to pursue a nursing career or advance your healthcare studies this comprehensive study guide is designed to help you excel From foundational knowledge to exam specific tips this guide provides the clarity and structured approach you need to tackle the HESI A2 with confidence This expertly crafted study guide covers all the essential subjects tested on the HESI A2 including Mathematics Reading Comprehension Grammar Vocabulary Biology Chemistry Anatomy and Physiology Physics and Critical Thinking Each chapter breaks down complex concepts into manageable sections providing clear explanations and detailed examples that ensure you understand the

material thoroughly With a focus on practical application this guide not only helps you grasp key principles but also trains you in effective test taking strategies It offers hundreds of practice questions with detailed answers and explanations to reinforce your knowledge ensuring you are well prepared for the actual exam You ll also find helpful tips for managing your time during the test and boosting your overall test performance Our approach ensures that you are not just memorizing facts but are gaining a deep understanding of the material Whether you need to brush up on basic math operations master reading comprehension techniques or sharpen your knowledge of human biology and chemistry this guide has you covered Additionally it provides insight into common pitfalls and how to avoid them ensuring that you feel prepared for even the most challenging questions The guide also includes a section on critical thinking and decision making skills which are vital for nursing and healthcare professions These skills will not only help you succeed on the HESI A2 exam but also throughout your nursing career Whether you re a first time test taker or looking to improve your score this guide offers everything you need to succeed With detailed explanations expert tips and a structured study plan it s the perfect resource to help you pass the HESI A2 exam and move one step closer to your dream of becoming a nurse Start your journey today and take the first step toward a rewarding healthcare career

watermaths Simon Judd ,2019-11-15 Watermaths presents the mathematics underpinning the design and operation of the individual unit process technologies used for purifying water and wastewater The book aims to provide the reader with sufficient information to enable them to tackle the most important calculations in this area without requiring any prior knowledge of the subject and assuming only a very basic grounding in science or engineering It focuses on the most essential areas of knowledge required containing tuition in basic numeracy chemistry process engineering and fluid physics as well as cost analysis The simple and succinct delivery is designed to get the reader up to speed as rapidly as possible sufficient background information is provided to explain the purpose of the calculations and ultimately tackle the complete wastewater reclamation plant design problem included in the book Example calculations are provided within each chapter each followed by exercises intended to reinforce the learning and for which solutions are appended Exercises range in difficulty from simple single calculational step problems to more complex ones and the overarching design problem provides some context to the mathematics The book can be understood by those relatively new to the water sector and is intended as a primer rather than a comprehensive handbook It is nonetheless sufficiently comprehensive to permit design calculations for most water and wastewater treatment unit processes Core disciplines covered include manipulation of equations including logarithmic and exponential expressions fluid physics for describing flow through pipes channels and filters chemical concentrations and chemical biochemical reactions chemical biochemical reaction kinetics mass balance for determining fate of materials through unit processes mass transfer for determining transfer of materials across boundaries within processes reactor theory for designing biochemical and chemical reaction vessels cost analysis including capital and operating expenditure with discounting New to the third edition new chapter on cost analysis further

explanation of the classical unit operations types illustrations expanded to include unit operation schematics and symbols
 new examples and exercises updated design problem Watermaths just add water **Chemistry** Trace Jordan, Neville R.
 Kallenbach, 2017 Chemistry The Molecules of Life offers chemical insights within the context of health pharmaceuticals and
 the function of biological molecules The contextualized presentation of topics gives students a broad introduction to
 chemistry and helps them to see the relevance of chemistry to their personal lives **Strategies and Solutions to
 Advanced Organic Reaction Mechanisms** Andrei Hent, John Andraos, 2019-06-28 Strategies and Solutions to Advanced
 Organic Reaction Mechanisms A New Perspective on McKillop's Problems builds upon Alexander Sandy McKillop's popular
 text Solutions to McKillop's Advanced Problems in Organic Reaction Mechanisms providing a unified methodological
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Problem-Solving Approach to Aquatic Chemistry James N. Jensen, 2023-01-05 A Problem Solving Approach to Aquatic Chemistry Enables civil and environmental engineers to understand the theory and application of aquatic equilibrium chemistry The second edition of A Problem Solving Approach to Aquatic Chemistry provides a detailed introduction to aquatic equilibrium chemistry calculation methods for systems at equilibrium applications of aquatic chemistry and chemical kinetics The text directly addresses two required ABET program outcomes in environmental engineering chemistry including stoichiometry equilibrium and kinetics and material and energy balances fate and transport of substances in and between air water and soil phases The book is very student centered with each chapter beginning with an introduction and ending with a summary that reviews the chapter's main points To aid in reader comprehension important terms are defined in context and key ideas are summarized Many thought provoking discussion questions worked examples and end of chapter problems are also included Each part of the text begins with a case study a portion of which is addressed in each subsequent chapter illustrating the principles of that chapter In addition each chapter has an Historical Note exploring connections with the people and cultures connected to topics in the text A Problem Solving Approach to Aquatic Chemistry includes Fundamental concepts such as concentration units thermodynamic basis of equilibrium and manipulating equilibria Solutions of chemical equilibrium problems including setting up the problems and algebraic graphical and computer solution techniques Acid base equilibria including the concepts of acids and bases titrations and alkalinity and acidity Complexation including metals ligands equilibrium calculations with complexes and applications of complexation chemistry Oxidation reduction equilibria including equilibrium calculations graphical approaches and applications Gas liquid and solid liquid equilibrium with expanded coverage of the effects of global climate change Other topics including chemical kinetics of aquatic systems surface chemistry and integrative case studies For advanced senior undergraduates and first year graduate students in environmental engineering courses A Problem Solving Approach to Aquatic Chemistry serves as an invaluable learning resource on the topic with a variety of helpful learning elements included throughout to ensure information retention and the ability to apply covered concepts in practical settings **Chemistry** Greg Curran, 2011 Covers all the topics in a typical one year high school chemistry curriculum **FCS Physical Science L3** Karen Morrison, 2008 *eBook: General, Organic and Biological Chemistry 2e* SMITH, 2012-02-16 *eBook General Organic and Biological Chemistry 2e* **Cehmistry Textbook for College and University USA** Ibrahim Sikder, 2023-06-04 Cehmistry Textbook USA **Longman Science Chemistry 10** Kohli Nitin, 2008-09 *Financial Management for Nurse Managers and Executives* Steven A. Finkler, Christine Tassone Kovner, Cheryl Bland Jones, 2007-01-01 Covering the financial topics all nurse managers need to know and use this book explains how financial management fits into the healthcare organization You'll study accounting principles cost analysis planning and control management of the organization's financial resources and the use of management tools In addition to

current issues this edition also addresses future directions in financial management Chapter goals and an introduction begin each chapter Each chapter ends with Implications For The Nurse Manager and Key Concepts to reinforce understanding Key Concepts include definitions of terms discussed in each chapter A comprehensive glossary with all key terms is available on companion Evolve website Two chapter ending appendixes offer additional samples to reinforce chapter content Four NEW chapters are included Quality Costs and Financing Revenue Budgeting Variance Analysis Examples Extensions and Caveats and Benchmarking Productivity and Cost Benefit and Cost Effectiveness Analysis The new Medicare prescription bill is covered with its meaning for healthcare providers managers and executives Coverage now includes the transition from the role of bedside or staff nurse to nurse manager and nurse executive Updated information includes current nursing workforce issues and recurring nursing shortages Updates focus on health financing and the use of computers in budgeting and finance New practice problems are included

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