

Summative Assessment: Nuclear Chemistry – Radioactivity & Nuclear Energy (Traditional Organization)

Matching: Match the description in column A with the correct term in column B. Write the letter in the blank provided. Each term matches with only one description, so be sure to choose the best description for each term.

Column A

- ___ **L** ___ 1. Used in the medical field to help provide information about how the body functions
- ___ **G** ___ 2. Helium nucleus
- ___ **C** ___ 3. Self-sustaining process
- ___ **M** ___ 4. Instrument used for measuring radioactivity levels
- ___ **D** ___ 5. Undergoes spontaneous nuclear decay
- ___ **I** ___ 6. High-energy photon
- ___ **A** ___ 7. Amount of radioactive material necessary for fission to occur
- ___ **H** ___ 8. Time it takes for half of a sample of radioactive nuclei to decay
- ___ **E** ___ 9. Splitting one heavy nucleus into two nuclei with smaller mass numbers
- ___ **N** ___ 10. Electron
- ___ **J** ___ 11. Particle with the same mass as an electron but a positive charge
- ___ **B** ___ 12. Subatomic particles found in the nucleus
- ___ **F** ___ 13. Combining two light nuclei to form a heavier nucleus
- ___ **K** ___ 14. The change of one element into another

Column B

- A. Critical Mass
- B. Nucleons
- C. Chain reaction
- D. Radioactive
- E. Fission
- F. Fusion
- G. Alpha particle
- H. Beta particle
- I. Gamma ray
- J. Positron
- K. Nuclear transmutation
- L. Radiotracers
- M. Geiger Counters
- N. Half-life

Multiple Choice - Choose the one best answer and write its letter in the left margin.

- ___ **C** ___ 15. Identify the missing particle: $^{239}_{94}\text{Pu} \rightarrow \text{_____} + ^4_2\text{He}$
- A. $^{235}_{92}\text{U}$
 B. $^{239}_{92}\text{Th}$
C. $^{235}_{92}\text{U}$
 D. $^{239}_{96}\text{Am}$

- ___ **B** ___ 16. Identify the missing particle: $^{238}_{90}\text{Th} \rightarrow \text{_____} + ^4_2\text{He}$

Unit 16 Nuclear Chemistry Answer Key

Mr. Ajeet Singh



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Table of Contents Unit 16 Nuclear Chemistry Answer Key

1. Understanding the eBook Unit 16 Nuclear Chemistry Answer Key
 - The Rise of Digital Reading Unit 16 Nuclear Chemistry Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Unit 16 Nuclear Chemistry Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Unit 16 Nuclear Chemistry Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Unit 16 Nuclear Chemistry Answer Key
 - Personalized Recommendations
 - Unit 16 Nuclear Chemistry Answer Key User Reviews and Ratings

- Unit 16 Nuclear Chemistry Answer Key and Bestseller Lists
- 5. Accessing Unit 16 Nuclear Chemistry Answer Key Free and Paid eBooks
 - Unit 16 Nuclear Chemistry Answer Key Public Domain eBooks
 - Unit 16 Nuclear Chemistry Answer Key eBook Subscription Services
 - Unit 16 Nuclear Chemistry Answer Key Budget-Friendly Options
- 6. Navigating Unit 16 Nuclear Chemistry Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Unit 16 Nuclear Chemistry Answer Key Compatibility with Devices
 - Unit 16 Nuclear Chemistry Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Unit 16 Nuclear Chemistry Answer Key
 - Highlighting and Note-Taking Unit 16 Nuclear Chemistry Answer Key
 - Interactive Elements Unit 16 Nuclear Chemistry Answer Key
- 8. Staying Engaged with Unit 16 Nuclear Chemistry Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Unit 16 Nuclear Chemistry Answer Key
- 9. Balancing eBooks and Physical Books Unit 16 Nuclear Chemistry Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Unit 16 Nuclear Chemistry Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Unit 16 Nuclear Chemistry Answer Key
 - Setting Reading Goals Unit 16 Nuclear Chemistry Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Unit 16 Nuclear Chemistry Answer Key
 - Fact-Checking eBook Content of Unit 16 Nuclear Chemistry Answer Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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