Chemistry Unit 8 Worksheet 4 Samples of Every Kind of Problem

On a separate sheet of paper, write a complete solution to each of the problems below. Follow the procedure outlined in class. Be sure to circle your final answer.

 Calculate the number of moles of potassium chlorate, KClO₃ (s), that must decompose to produce potassium chloride, KCl (s), and 1.8 moles of oxygen gas.

Equation:	2 KClO ₃ →	2 KCI +	3 O2
Before:	1.2	0	0
Change:	-1.2	+1.2	+1.8
After:	0	1.2	1.8

1.2 moles KClO₃ needed

2. In a single displacement reaction, magnesium metal reacts with hydrochloric acid to produce magnesium chloride and hydrogen gas. How many moles of hydrochloric acid are needed to completely react with 2.43 g of magnesium?

Equation:	Mig +	2 HCl ->	MgCl ₂ +	H ₂
Before:	0.100	- 7	0	0
Change:	-0.100	-0.200 +	0.160 +	0.100
Afters	0		0.100	0.100

0,200 moles HCI needed

0.73 g x 1 mole = .0040 mole 183 g

3. Ethane, C₂H₆ reacts with oxygen gas to produce carbon dioxide gas and water vapor. What mass of oxygen gas is required to react with 2.20 moles of ethane?

Equation:	2 C2H6	+ 7 O ₂	→ 4 CO₂	+ 6 H ₂ O
Before:	2.20	36.56	0	0
Change:	-2.20	- 7.70	+ 4,40	+ 6.60
After:	0	X 5	4.40	6.60

O₂ needed: 7.70 mole x 32.0 g = 246 g O₂ required 1 mole

 Determine the mass of sodium nitrate produced when 0.73 g of nickel (II) nitrate reacts with sodium hydroxide according to the following equation: Ni(NO3)2 + 2 NaOH --> Ni(OH)2 + 2 NaNO3

 Equation:
 Ni(NO₃)₂ + 2 NaOH
 → Ni(OH)₂ + 2 NaNO₃

 Before:
 0.0040
 XS
 0
 0

 Change:
 -0.0040
 -0.0080
 +0.0040
 +0.0080

 After:
 0
 XS
 0.0040
 0.0080

NaNO, produced: 0.0080 mole x 85.0 g = 0.68 g 1 mole

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counterion nature concentration of starting materials presence of electron donors and redox potentials The behavior of macroions 2nm 6nm size range in solution is shown to be quite different from the simple ionic solution or colloidal systems DebyeOCoHuckel model Their self assembling into a single layered spherical hollow vesicle structure namely the OC blackberryOCO structure is clearly described Examples of spin clusters with tunable interactions are given and single molecule magnets based on POMs are specifically tackled Besides paramagnetic transition metal centres and lanthanoid ions encapsulated in archetypal lacunary polyoxoanions magnetically functionalized Kleperates are described their discovery Catalogue of Marshall College, State Normal School, Huntington, W. Va Marshall tracing back nearly 15 years College (Huntington, W. Va.), Marshall University, 1914 Bulletin University of Minnesota, 1919 Announcement of Courses Stanford University, 1912 Register ... California. University, University of California, Berkeley, 1914 Catalogue of the College of California and College School University of California (System),1913 **Annual Register of the State** University of Nevada ... with Announcements ... University of Nevada, 1924

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