

Unit: Chemical Bonding Answer Key "Covalent Bonding" - WS #3

Directions: Please answer each question using your notes from today's class.

©

 What is the difference between a molecular formula, structural formula and an electron dot formula? Give an example of each.

A molecular formula gives the chemical make-up of one molecule. It does not show atom-to-atom connectivity. Example: C_2H_2

A structural formula uses lines instead of dots to show shared pairs of electrons.

Example: H-H

An electron dot formula uses dots instead of lines to show shared pairs of electrons. Example: H:H

2. What is the difference between single, double, and triple covalent bonds? How many shared electrons are in each?

The difference between the types of bonds is: a single bond is two shared electrons, a double is four shared electrons, and a triple is six shared electrons.

Draw the electron dot and structural formulas for the following molecules.

Substance	Rough copy (work!)	Electron dot	Structural
a) Cl ₂		Note: The electron dot structures have dots in between each atom instead of lines.	:çı—çı:
b) H ₂ S			:\$-н Н
c) Brl			ij <u> </u>

Unit 4 Lesson 20 Covalent Bonds Packet Answers

Jin-Ying Zhang

Unit 4 Lesson 20 Covalent Bonds Packet Answers: Books in Print Supplement ,2002

Unit 4 Lesson 20 Covalent Bonds Packet Answers Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has be apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Unit 4 Lesson 20 Covalent Bonds Packet Answers**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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