

Unisa provisional examination timetable for January 2012

Select the study units for which you wish to register in such a manner that the examination dates and times do not clash.

NB: No examination clashes will be permitted. Public holidays in foreign countries are not taken into account when compiling the examination timetable.

Code	Paper	Date	Time
A			
AAR401F	1	2012/01/09	14:30
AAR402G	1	2012/01/11	14:30
AAR403H	1	2012/01/13	14:30
AAR404J	1	2012/01/15	14:30
AAS405K	1	2012/01/10	08:30
ABT101B	1	2012/01/11	11:30
ABT102C	1	2012/01/12	08:30
ABT103D	1	2012/01/13	11:30
ABT104E	1	Dept requirements	-
ABT110Z	1	2012/01/11	11:30
ABT110Z	1	2012/01/12	08:30
ABT110Z	1	2012/01/13	11:30
ABT110Z	1	2012/01/14	14:30
ABT110Z	1	2012/01/15	17:30
ABT110Z	1	2012/01/16	20:30
ABT110Z	1	2012/01/17	23:30
ABT110Z	1	2012/01/18	02:30
ABT110Z	1	2012/01/19	05:30
ABT110Z	1	2012/01/20	08:30
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ABT110Z	1	2012/03/27	17:30
ABT110Z	1	2012/03/28	20:30
ABT110Z	1	2012/03/29	23:30
ABT110Z	1	2012/03/30	02:30
ABT110Z	1	2012/03/31	05:30

Code	Paper	Date	Time
ADB401M	1	2012/01/13	14:30
ADB401M	1	2012/01/15	14:30
ADD523P	1	Dept requirements	-
ADP346V	1	Dept requirements	-
ADP400E	1	2012/01/17	14:30
ADVCATA	1	Dept requirements	-
ADV401F	1	2012/01/12	08:30
ADV401F	1	2012/01/13	11:30
ADV401F	1	2012/01/14	14:30
ADV401F	1	2012/01/15	17:30
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ADV401F	1	2012/03/29	23:30
ADV401F	1	2012/03/30	02:30
ADV401F	1	2012/03/31	05:30

Code	Paper	Date	Time
AND301A	1	2012/01/13	11:30
AND301D	1	2012/01/15	08:30
AND401D	1	2012/01/09	14:30
ANE400Z	1	2012/01/16	14:30
ANE400Z	1	2012/01/18	14:30
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NB: Please note that the dates reflected in these timetables are provisional and the University reserves the right to change examination dates if necessary.

Unisa Provisional Timetable For 2015

Huangqi Zhang



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Whispering the Techniques of Language: An Emotional Journey through **Unisa Provisional Timetable For 2015**

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web jan 28 2019 by mapping specific trends directly onto a blank periodic table the activity uses the periodic table as a foundational aspect of the major trends allowing for students to make connections between placement on the periodic table

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web you might need periodic table the first two ionization energies for beryllium are shown below be g be g e be g be 2 g e i 1 900 kJ mol i 2 1757 kJ mol which of the following identifies the most probable value for the third ionization energy for be and provides the best justification

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web the purpose of this lab is to introduce students to periodic trends it gives students an opportunity to explore the values of different properties for all elements in the periodic table to determine which properties of the elements follow a periodic trend

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web 1 how does atomic radius change as you move from left to right across a period explain your answer in terms of the number of protons in the nucleus 2 how does atomic radius change as you move from top to bottom in a group explain your answer in terms of the number of electron shells 3 on the representation of the periodic table below

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