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Date Pd

Unit 3 Worksheet 4 – Quantitative Energy Problems Part 2

Energy constants (H2O)

334 J/g Heat of fusion (melting or freezing) H_f

2260 J/g Heat of vaporization (evaporating or condensing) H_V

2.1 J/g°C Heat capacity (c) of solid water 4.18 J/g°C Heat capacity (c) of liquid water

For each of the problems sketch a warming or cooling curve to help you decide which equation(s) to use to solve the problem. Keep a reasonable number of sig figs in your answers.

 How much energy must be absorbed by a 150 g sample of ice at 0.0 °C that melts and then warms to 25.0 °C?

Suppose in the ley Hot lab that the burner transfers 325 kJ of energy to 450 g of liquid water at 20. °C. What mass of the water would be boiled away?

3. A 12oz can of soft drink (assume m = 340 g) at 25°C is placed in a freezer where the temperature is - 12 °C. How much energy must be removed from the soft drink for it to reach this temperature?

Unit Worksheet Quantitative Energy Problems

United States. National Bureau of Standards

Unit Worksheet Quantitative Energy Problems:

Quantitative Reasoning in the Context of Energy and Environment Robert Mayes, James Myers, 2015-01-19 This book provides professional development leaders and teachers with a framework for integrating authentic real world performance tasks into science technology engineering and mathematics STEM classrooms We incorporate elements of problem based learning to engage students around grand challenges in energy and environment place based leaning to motivate students by relating the problem to their community and Understanding by Design to ensure that understanding key concepts in STEM is the outcome Our framework has as a basic tenet interdisciplinary STEM approaches to studying real world problems We invited professional learning communities of science and mathematics teachers to bring multiple lenses to the study of these problems including the sciences of biology chemistry earth systems and physics technology through data collection tools and computational science modeling approaches engineering design around how to collect data and mathematics through quantitative reasoning Our goal was to have teachers create opportunities for their students to engage in real world problems impacting their place problems that could be related to STEM grand challenges demonstrating the importance and utility of STEM We want to broaden the participation of students in STEM which both increases the future STEM workforce providing our next generation of scientists technologists engineers and mathematicians as well as producing a STEM literate citizenry that can make informed decisions about grand challenges that will be facing their generation While we provide a specific example of an interdisciplinary STEM module we hope to do more than provide a single fish Rather we hope to teach you how to fish so you can create modules that will excite your students **Energy Abstracts for Policy Analysis** ,1984-07

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