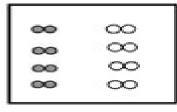
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# Chemistry Unit 8 Worksheet 3: Adjusting to Reality - Limiting Reactant

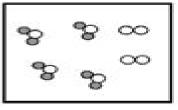
Write the balanced equation for the reaction between hydrogen and oxygen.

Balanced Equation:  $2 H_{2(g)} + 1 O_{2(g)} \rightarrow 2 H_2O_{(g)}$ Suppose that 4 molecules of hydrogen gas and 4 molecules of oxygen gas react to form water.

Make a drawing that represents the reaction container before and after the reaction.







Before

After

4 molecules How many molecules of water can be produced?

Oxygen Which reactant is in excess? Why? There are leftover molecules

2 molecules How many molecules of excess reactant are there?

Construct a Before-Change-After Table for this reactant mixture: In this table the numbers refer to molecules rather than moles. Equation:  $2 H_{2(g)} + 1 O_{2(g)} \rightarrow 2 H_{2}O_{(g)}$ Before  $4 U_{2(g)} \rightarrow 2 U_{2(g)} \rightarrow 2 U_{2(g)} \rightarrow 2 U_{2(g)}$ 

Change -1 -2 +4
After 0 2. 4

According to the table you just made,

4 molecules
Oxygen
Which reactant is in excess? Why? 2 molecules are left-over "after" the reaction

2 molecules
How many molecules of excess reactant are there?

Based on your two methods of analysis above, what determines how much product can be made from a particular reactant mix?

The determining factor is which of the reactants runs out first. This determines how much product can be made, even if another reactant is in excess.

# **Unit 7 3 Limiting Reactant Answers**

Judd E. Hollander

Unit	7	3	Limiting	Reactant	Answers:
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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Unit 7 3 Limiting Reactant Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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