

SCIENCE 8 – DENSITY CALCULATIONS WORKSHEET

NAME: Key

- 1) A student measures the mass of an 8 cm^3 block of brown sugar to be 12.9 g . What is the density of the brown sugar?

$$m = 12.9 \text{ g}$$

$$V = 8 \text{ cm}^3$$

$$D = ?$$

$$D = \frac{m}{V}$$

$$D = \frac{12.9 \text{ g}}{8 \text{ cm}^3}$$

$$D = 1.6125 \text{ g/cm}^3$$

- 2) A chef fills a 50 mL container with 43.5 g of cooking oil. What is the density of the oil?

$$V = 50 \text{ mL}$$

$$m = 43.5 \text{ g}$$

$$D = \frac{m}{V}$$

$$D = \frac{43.5 \text{ g}}{50 \text{ mL}}$$

$$D = .87 \text{ g/mL}$$

- 3) Calculate the mass of a liquid with a density of 2.5 g/mL and a volume of 15 mL .

$$D = 2.5 \text{ g/mL}$$

$$V = 15 \text{ mL}$$

$$2.5 \text{ g/mL} = \frac{m}{15 \text{ mL}}$$

$$m = (2.5)(15)$$

$$m = 37.5 \text{ g}$$

- 4) Calculate the volume of a liquid with a density of 5.45 g/mL and a mass of 65 g .

$$D = 5.45 \text{ g/mL}$$

$$m = 65 \text{ g}$$

$$5.45 = \frac{65}{V}$$

$$V = \frac{65}{5.45}$$

$$V = 11.93 \text{ mL}$$

- 5) A machine shop worker records the mass of an aluminum cube as 176 g . If one side of the cube measures 4 cm , what is the density of the aluminum?

$$m = 176 \text{ g}$$

$$V = L \times W \times H$$

$$D = \frac{176 \text{ g}}{64 \text{ cm}^3}$$

$$D = 2.75 \text{ g/cm}^3$$

- 6) A scientist performing a demonstration finds that a piece of cork displaces 23.8 mL of water. The piece of cork has a mass of 5.8 g . What is the density of the cork?

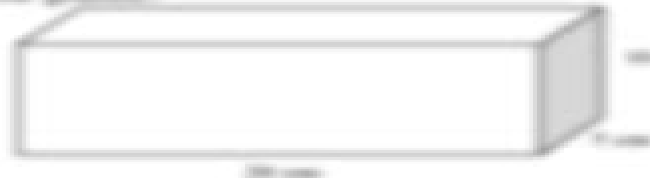
$$m = 5.8 \text{ g}$$

$$V = 23.8 \text{ mL}$$

$$D = \frac{5.8}{23.8}$$

$$D = .24 \text{ g/mL}$$

- 7) A carpenter begins work on the following block of granite that weighs 2700 g . What is the density of the granite?



$$m = 2700 \text{ g}$$

$$V = L \times W \times H$$

$$= 20 \times 10 \times 8$$

$$= 1600 \text{ cm}^3$$

$$D = \frac{2700}{1600}$$

$$D = 2.7 \text{ g/cm}^3$$

- 8) A piece of PVC plumbing pipe displaces 48 mL when placed into a container of water. If the pipe has a mass of 78 g , what is the density of PVC?

$$V = 48 \text{ mL}$$

$$m = 78 \text{ g}$$

$$D = \frac{78}{48}$$

$$D = 1.63 \text{ g/mL}$$

- 9) A solid magnesium flake has a mass of 1700 g and a volume of 743 cm^3 . What is the density of the magnesium?

$$m = 1700 \text{ g}$$

$$V = 743 \text{ cm}^3$$

$$D = \frac{1700}{743}$$

$$D = 2.29 \text{ g/cm}^3$$

Unit 3 Measurements And Calculations Answer Key

**Allan Blackman, Daniel
Southam, Gwendolyn Lawrie, Natalie
Williamson, Christopher Thompson**

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Chemistry Greg Curran, 2011 Covers all the topics in a typical one year high school chemistry curriculum Jacaranda Chemistry 2 VCE Units 3 and 4, 3e learnON and Print Robert Stokes, Angela Stubbs, Neale Taylor, Jason Bourke, Ben Williams, Maida Derbogosian, 2023-12-26 Developed by expert Victorian teachers for VCE students The NEW Jacaranda Chemistry VCE series continues to deliver curriculum aligned material that caters to students of all abilities Our expert author team of practising teachers and assessors ensures 100% coverage of the new VCE Chemistry Study Design 2023 2027

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Cambridge IGCSE Core Mathematics Practice Book Karen Morrison, Lucille Dunne, 2012-09-06 A series of titles written to cover the complete Cambridge IGCSE Mathematics 0580 syllabus and endorsed by Cambridge International Examinations

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