

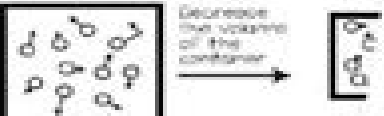
GAS LAWS

Name: _____

Gas molecules are constantly in motion and of low mass and the walls of the container. For each of, write whether the number of collisions per minute increases and then explain your answer.

2.  Double the number of molecules.

Collisions per minute will _____ (increase/decrease)

3.  Decrease the volume of the container.

Collisions per minute will _____ (increase/decrease)

4.  Increase the temperature.

Collisions per minute will _____ (increase)

Name: _____

Answer the questions below. Use the space below each set of questions for your work.

Class: _____

Date: _____

Charles' Law

A fun-loving chemist has a balloon at standard temperature and pressure and the balloon has a volume of 3.00 L.



1. What is the initial volume of the balloon?

2. What is the external temperature?

3. If the pressure is held constant and the balloon is heated to 40.0°C, what will the new volume of the balloon be?

4. What is the new temperature of the system?

5. What is the new volume of the balloon?



From the fun-loving chemist - placed in a box of air.

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Albert A Gayle



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