

Chemistry – Unit 5 – Worksheet 2

Show work. Label set-ups and answers with units. *Box answers.*

1. An old (pre-1987) penny is nearly pure copper. If such a penny has a mass of 3.3 g, how many moles of copper atoms would be in one penny?
2. Four nails have a total mass of 4.42 grams. How many moles of iron atoms do they contain?
3. What mass of water would you need to have 15.0 moles of H_2O ?
4. One box of Morton's Salt contains 737 grams. How many moles of sodium chloride, NaCl , is this?
5. A chocolate chip cookie recipe calls for 0.050 moles of baking soda (sodium bicarbonate). Sodium bicarbonate has the chemical formula, NaHCO_3 . How many grams should the chef mass out?
6. Lead (II) chromate, PbCrO_4 , was used as a pigment in paints. How many moles of PbCrO_4 are in 75.0 g of lead (II) chromate? How many atoms of oxygen are present?

Unit 2 Smells Chemistry Worksheets

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Table of Contents Unit 2 Smells Chemistry Worksheets

1. Understanding the eBook Unit 2 Smells Chemistry Worksheets
 - The Rise of Digital Reading Unit 2 Smells Chemistry Worksheets
 - Advantages of eBooks Over Traditional Books
2. Identifying Unit 2 Smells Chemistry Worksheets
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Unit 2 Smells Chemistry Worksheets
 - User-Friendly Interface
4. Exploring eBook Recommendations from Unit 2 Smells Chemistry Worksheets
 - Personalized Recommendations
 - Unit 2 Smells Chemistry Worksheets User Reviews and Ratings
 - Unit 2 Smells Chemistry Worksheets and Bestseller Lists
5. Accessing Unit 2 Smells Chemistry Worksheets Free and Paid eBooks
 - Unit 2 Smells Chemistry Worksheets Public Domain eBooks

- Unit 2 Smells Chemistry Worksheets eBook Subscription Services
- Unit 2 Smells Chemistry Worksheets Budget-Friendly Options
- 6. Navigating Unit 2 Smells Chemistry Worksheets eBook Formats
 - ePub, PDF, MOBI, and More
 - Unit 2 Smells Chemistry Worksheets Compatibility with Devices
 - Unit 2 Smells Chemistry Worksheets Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Unit 2 Smells Chemistry Worksheets
 - Highlighting and Note-Taking Unit 2 Smells Chemistry Worksheets
 - Interactive Elements Unit 2 Smells Chemistry Worksheets
- 8. Staying Engaged with Unit 2 Smells Chemistry Worksheets
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Unit 2 Smells Chemistry Worksheets
- 9. Balancing eBooks and Physical Books Unit 2 Smells Chemistry Worksheets
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Unit 2 Smells Chemistry Worksheets
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Unit 2 Smells Chemistry Worksheets
 - Setting Reading Goals Unit 2 Smells Chemistry Worksheets
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Unit 2 Smells Chemistry Worksheets
 - Fact-Checking eBook Content of Unit 2 Smells Chemistry Worksheets
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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