Balancing Act

Name

Atoms are not <u>created</u> or <u>destroyed</u> during a chemical reaction.

Scientists know that there must be the <u>same</u> number of atoms on each <u>side</u> of the <u>reaction</u>. To balance the chemical equation, you must add <u>coefficients</u> in front of the chemical formulas in the equation. You cannot <u>add</u> or <u>remove</u> subscripts.

- 1) Determine number of atoms for each element
- 2) Pick an element that is not equal on both sides of the equation.
- 3) Add coefficients in front of the element and adjust your counts.
- Continue adding coefficients until you get the same number of elements on each side.

Example:

$$2$$
 Mg + O_2 \rightarrow 2 MgO
Mg = 2 Mg = 2
 $O = 2$ $O = 2$

Try these:

$$N_2$$
 + 3 H_2 \rightarrow 2 NH_3
 $N = 2$
 $H = 6$ $H = 6$

$$2 H_2O_2 \rightarrow 2 H_2O + O_2$$
 $H = 4 H = 4$
 $O = 4 O = 4$

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Daniela Niemeyer

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groups are equal or not. Learn more by following along with our ... 1.3.5.3. Two-Sample <i>t</i>-Test for Equal Means Purpose: Test if two population means are equal, The two-sample t-test (Snedecor and Cochran, 1989) is used to determine if two population means are equal. 2 Sample t-Test (1 tailed) Suppose we have two samples of ceramic sherd thickness collected from an archaeological site, where the two samples are easily distinguishable by the use of. Two sample t-test: SAS instruction Note that the test is two-sided (sides=2), the significance level is 0.05, and the test is to compare the difference between two means (mu1 - mu2) against 0 (h0 ...