

Program: B.Tech

Course Code &amp; Title: 21CVB101J/Chemistry

Year &amp; Sem: I/II

Date: 29-03-2023

Time: 12.30-1.30 pm

Max. Marks: 30 Marks

Answer ALL the MCQs Part-A (10\*1=10 Marks)

- Which of the following is not a state function  
a)  $\Delta G$  b)  $\Delta E$  c)  $W$  d)  $H$
- Internal energy does not include  
a) Nuclear energy b) Vibrational energy c) Rotational energy d) Energy of gravitational pull
- Gibbs energy change  $\Delta G$  is related to equilibrium constant  $K$  as:  
a)  $\Delta G^\circ = -RT \ln k$  b)  $\Delta G^\circ = RT \ln k$  c)  $\ln k = -RT/\Delta G^\circ$  d)  $\ln k = \Delta G^\circ/RT$
- The conductivity of electrolytic conductors is due to \_\_\_\_\_  
a) Flow of free mobile electrons b) Movement of ions c) Either movement of electrons or ions d) Cannot be said
- The process of deterioration of a metal due to unwanted chemical or electrochemical interaction of the metal with its environment is called \_\_\_\_\_  
a) Electrolysis b) Electrodialysis c) Corrosion d) Deposition
- Chemical used in salt bridge is  
a)  $KOH$  b)  $KCl$  c)  $KNO_3$  d)  $KBr$
- If a compound has 3 chiral carbons What is the number of optically active isomers?  
a) 9 b) 3 c) 4 d) 8
- The potential energy of n-butane is minimum for  
a) Skew conformations b) Staggered conformations c) Eclipsed conformations d) Gauche
- Optical isomerism is a type of \_\_\_\_\_  
a) metamerism b) stereoisomerism c) geometrical isomerism d) tautomerism

# Srm University Dspace Previous Question Papers

**Terry C. Jones**



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