

SQA N5 Scotland Chemistry paper 2 2025

(i) Calculate the average rate of reaction, in $\text{cm}^3 \text{s}^{-1}$, for the first 50 seconds. 2

Show your working.

$$R = \frac{\Delta \text{time}}{\Delta \text{amount}}$$

$\Delta \text{time} = 0 + 85 \text{ cm}^3$

Ans. $1.7 \text{ cm}^3 \text{s}^{-1}$

(ii) Add a curve to the graph to show the effect of the experiment. 1

(An additional graph is provided.)

line

(iii) A similar experiment was carried out using zinc powder and the same volume of sulfuric acid. Predict the volume of gas produced. 1



* X 8 1 3 7 5 0 1 0 7 *

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 - The Rise of Digital Reading Sqa 2higher Chemistry Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying Sqa 2higher Chemistry Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Sqa 2higher Chemistry Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Sqa 2higher Chemistry Paper
 - Personalized Recommendations
 - Sqa 2higher Chemistry Paper User Reviews and Ratings
 - Sqa 2higher Chemistry Paper and Bestseller Lists
5. Accessing Sqa 2higher Chemistry Paper Free and Paid eBooks

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 - Sqa 2higher Chemistry Paper eBook Subscription Services
 - Sqa 2higher Chemistry Paper Budget-Friendly Options
6. Navigating Sqa 2higher Chemistry Paper eBook Formats
- ePub, PDF, MOBI, and More
 - Sqa 2higher Chemistry Paper Compatibility with Devices
 - Sqa 2higher Chemistry Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Sqa 2higher Chemistry Paper
 - Highlighting and Note-Taking Sqa 2higher Chemistry Paper
 - Interactive Elements Sqa 2higher Chemistry Paper
8. Staying Engaged with Sqa 2higher Chemistry Paper
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Sqa 2higher Chemistry Paper
9. Balancing eBooks and Physical Books Sqa 2higher Chemistry Paper
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Sqa 2higher Chemistry Paper
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Sqa 2higher Chemistry Paper
- Setting Reading Goals Sqa 2higher Chemistry Paper
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Sqa 2higher Chemistry Paper
- Fact-Checking eBook Content of Sqa 2higher Chemistry Paper
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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