

Experiment 35**Advance Study Assignments: Spot Tests for Some Common Anions**

1. Each of the observations in the following list was made on a different solution. Given the observations, state which ion studied in this experiment is present. If the test is not definitive, indicate that with a question mark.
 - A. Addition of 6 M NaOH and Al to the solution produces a vapor that turns red litmus blue.
Ion present: _____
 - B. Addition of 6 M HCl produces a vapor with an acid odor.
Ion present: _____
 - C. Addition of 6 M HCl produces an effervescent.
Ion present: _____
 - D. Addition of 6 M HNO₃, plus 0.1 M AgNO₃, produces a precipitate.
Ion present: _____
 - E. Addition of 6 M HNO₃, plus 1 M BaCl₂, produces a precipitate.
Ion present: _____
 - F. Addition of 6 M HNO₃, plus 0.5 M (NH₄)₂MoO₄, produces a precipitate.
Ion present: _____
2. An unknown containing one or more of the ions studied in this experiment has the following properties:
 - A. No effect is observed on addition of 6 M HNO₃.
 - B. No effect is observed on addition of 0.1 M AgNO₃ to the solution from Part A.
 - C. A white precipitate is observed on addition of 1 M BaCl₂ to the solution from Part A.
 - D. A yellow precipitate is observed on addition of (NH₄)₂MoO₄ to the solution from Part A.

On the basis of this information, which ions are present, which are absent, and which are in doubt?

Present

Absent

In doubt

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web learn what it means for two figures to be similar and how to determine whether two figures are similar or not use this
concept to prove geometric theorems and solve some problems with polygons

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web teaching students all of the facets of similar shapes in geometry can be a challenge when i try to do it in one day so this
worksheet is intended to take a little pressure off of me and put the information and work into the students hands the
worksheet has them identify similar shapes explaini