	-2: Writing Balanced Precipitation Reactions Lab che	emistry
si	this problem, you will go into the virtual laboratory and perform a series of precipitatio ing Ag*, Pb2*, and Sb3*. After observing the reactions, you will write the net ionic equat presenting these reactions and then balance them.	
	Start Virtual ChemLab, select Reactions and Stoichiometry, and then select Writing Be Precipitation Reactions from the list of assignments. The lab will open in the Inorgani	
2.	Drag a test tube from the box and place it on the metal test tube stand. You can then cl bottle of Ag* ion solution (it is AgNO ₃) on the shelf to add it to the test tube. Then dra to the blue test tube rack. Repeat for Pb ^{2*} (Pb(NO ₃) ₂ and Sb ^{3*} (Sb(OH) ₃).	
3.	Moving your mouse over each test tube in the test-tube rack will allow you to identify monitor. You can also click on the label at the top of each test tube in order to label the Place the test tube containing the Ag* solution in the metal test tube stand. You will be observe what happens in the window at the top left. Click on the Na ₂ CO ₃ reagent bottle the test tube in the stand.	e test tubes.
	What color is the precipitate? Write out the full balanced molecular equation for the re	eaction.
	Write the balanced ionic equation for the reaction.	

<u>Virtual Lab Writing Balanced Precipitation Reactions</u> <u>Answers</u>

Ralph Petrucci, William Harwood, Geoffrey Herring

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