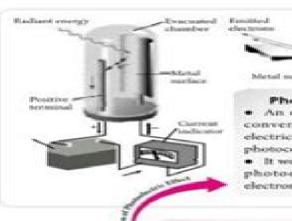
# QUANTUM THEORY OF LIGHT





# Martial surrisco

#### Photoelectric Cell

- · An electrical device which converts light energy into electrical energy, is called as photocell or photoelectric cell.
- · It works on the principle of photoelectric emission of electrons.

#### Photoelectric Effect

· The phenomenon of emission of electrons from a metal surface when an electromagnetic wave of suitable frequency is incident on it is called photoelectric officet.

#### **Basic Quantum Theory of Light**

According to Planck, the energy of a photon, E = 0;

$$E = hv_0 = \frac{hc}{\lambda} = \frac{1240}{\lambda (\ln nm)} eV$$

- Momentum of photon,  $p = \frac{E}{n} = \frac{hv}{n} = \frac{h}{n}$
- If source is 100% efficient, then the number of photons emitted per second by the source can be given by

$$= \frac{\text{Power of source}}{\text{Energy of photon}} = \frac{P}{E} = \frac{P}{hv} = \frac{P\lambda}{hc}$$

- The energy crossing per unit area per unit time perpendicular to the direction of propagation is called the intensity of a wave. I = E / A E = P / A
- · Force exerted on perfectly reflecting surface

$$F = \frac{\Delta p}{t} = \frac{2Nh}{t\lambda} = n\left(\frac{2h}{\lambda}\right) = \frac{2P}{c}$$

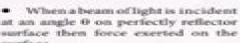
$$F = 2P = 2I$$

Pressure = 
$$\frac{F}{A} = \frac{2P}{cA} = \frac{2I}{c}$$

Force exerted on perfectly absorbing surface

$$F = \frac{\Delta p}{t} = \frac{Nh}{t\lambda} = n\left(\frac{h}{\lambda}\right) = \frac{p}{c}$$

Pressure = 
$$\frac{F}{A} - \frac{P}{cA} - \frac{I}{c}$$



surface,  

$$F = \frac{2P}{\epsilon} \cos \theta = \frac{2IA \cos \theta}{\epsilon}$$
Pressure =  $\frac{2I \cos \theta}{\epsilon}$ 





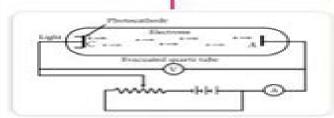


#### Photoelectric Equation

•  $E = K_{max} + \Phi_n$ where & = work function of metal. E = energy of incident light, $K_{max} = maximum kinetic energy of electrons$ 

Principal Services (Services)

 $=\frac{1}{2}mv_{state}^2 = h(\upsilon - \upsilon_0) = hc\left(\frac{1}{\lambda} - \frac{1}{\lambda_{\infty}}\right)$ where,  $\lambda_0 = \frac{hc}{\lambda}$  = threshold wavelength



#### Conclusions of Experimental Study of Photoelectric Effect

PARTICLE

NATURE OF

RADIATION

Photo-current is directly proportional to the ... intensity of incident light, i.e.,  $i_n \approx I$ .

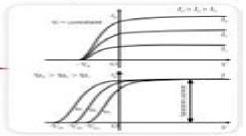
(At constant frequency trand potential V)

- · At constant frequency and intensity, the minimum negative potential at which the photocurrent becomes zero is called stopping protential  $(V_n)$ .
- At stopping potential V<sub>0</sub>, K<sub>max</sub> = eV<sub>0</sub>
- · For a given frequency of the incident radiation, the stopping potential is independent of its internsity.
- · The stopping potential varies linearly with the frequency of incident radiation but saturation current value remains constant for a fixed intensity of incident radiation.

Quantum efficiency

n, = number of electron emitted persecond

n<sub>ph</sub> = total number of photon incident per second



# **Study Guide Quantum Theory**

**S Ashworth** 

**Study Guide Quantum Theory:** 

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