Worksheet 6.2 Word Equations

- Write the chemical equations and balance each of the following word equations.
 - Aluminum metal reacts with iron (ii) oxide powder to produce aluminum oxide solid and iron metal.

LAI +3FED -7 A1,0,+3FE

 Aluminum sulphate solution and calcium hydroxide solution produce a precipitate of aluminum hydroxide and solid calcium sulphate.

AL(50+) 3 +3 Ca(OH) = - 72AKOH) + 36aBOH

Ammonia gas (NH₂) plus oxygen gas yields nitrogen monoxide gas plus water vapour.

4NH, +10, -74NO + (H10

 Calcium hydroxide solution and carbon dioxide gas yields solid calcium carbonate and liquid water.

(a(0H)) + CO2 -> (a(60))+ HIO

 Aqueous Iron (III) chloride and sodium carbonate solution yields aqueous sodium chloride and a precipitate of iron (III) carbonate.

2 Fellopes +) Markey ->6No (1 + Fc, (10))

f) Solid iron (III) axide and carbon monoxide gas yields iron metal and carbon dioxide gas.

Fe202 +260 -> 2Fe + 2602

Magnesium carbonate solution plus aqueous hydrochloric acid (HCI) yields magnesium chloride solution plus liquid water and carbon diexide gas.

My cos + 2HCI - Mycle + 400 + 600

 Silicon diaxide solid plus aqueous hydrofluoric acid (HF) yields solid silicon tetrafluoride plus liquid water.

5,020+44FGgs -> SiF46) +2H10

 Aqueous sodium hydroxide and carbon dioxide gas yields sodium carbonate solution and liquid water.

INOH + CO2 -7 NagCoz + HID

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Andrew Madden

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Writing Chemical Equations and Identifying Unknown Solutions Using Microscale Techniques Marcia L. Gillette, H. Anthony Neidig, 1992-01-01 Chemistry Equations & Answers Speedy Publishing, 2014-08-01 One of the most important parts of learning chemistry is simply knowing all of the equations and formulas that are used in it It is important to know the exact formatting of these equations and most classes will require a student to know them for exams It is a good idea to learn these equations and formulas with the use of a study pamphlet The pamphlet can condense all of the information so a student can memorize the equations and formulas while studying Writing Chemical & Quations Marcia L. Gillette, 1993-01-01

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