

Writing Chemical Equations

1. Identify reactants and products and place them in a word equation.
2. Convert the chemical names into chemical formulas and write the state symbols.
3. Balance the chemical equation.

Example:

Step 1: Aluminum + iron(III) oxide $\Rightarrow$ aluminum oxide + iron

Step 2: $\text{Al}_{(s)}$ $\text{Fe}_2\text{O}_{3(s)}$ $\text{Al}_2\text{O}_{3(s)}$ $\text{Fe}_{(l)}$

Step 3: $2\text{Al}_{(s)} + \text{Fe}_2\text{O}_{3(s)} \Rightarrow \text{Al}_2\text{O}_{3(s)} + 2\text{Fe}_{(l)}$

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