Worksheet 6.2 Word Equations

- Write the chemical equations and balance each of the following word equations.
 - Aluminum metal reacts with iron (II) oxide powder to produce aluminum oxide solid and iron metal.

1AI +3FED -> A1,0,+7FE

 Aluminum sulphate solution and calcium hydroxide solution produce a precipitate of aluminum hydroxide and solid calcium sulphate.

AL(50+) 3 +3 (. (OH), -> LAKOH) + 3 CaBOJ

Ammonia gas (NH_{*}) plus oxygen gas yields nitrogen monoxide gas plus water vapour.

4NH, +102 -7 4ND + (H20

 Calcium hydroxide solution and carbon dioxide gas yields solid calcium carbonate and liquid water.

[a(OH), + CO, -> (a(CO))+ HIO

 Aqueous iron (III) chloride and sodium carbonate solution yields aqueous sodium chloride and a precipitate of iron (III) carbonate.

2 Fe Charles +) Nos (cus) -> 6No (1 + Fc, (cos).

Solid iron (III) oxide and carbon monoxide gas yields iron metal and carbon dioxide gas.

Fe202 +260 -> 2Fe + 2002

 Magnesium carbonate solution plus aqueous hydrochloric acid (HCI) yields magnesium chloride solution plus liquid water and carbon dioxide gas.

Ma cos + 2401 - Macle + 400 + cos

 Silicon dioxide solid plus aqueous hydrofluoric acid (HF) yields solid silicon tetrafluoride plus liquid water.

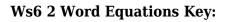
5,020+44Flags -5 5; F46) +2H10

 Aqueous sodium hydroxide and carbon dioxide gas yields sodium carbonate solution and liquid water.

2 NOH + CO2 -7 Nag CO2 + HO

Ws6 2 Word Equations Key

Annelies Wilder-Smith



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