

## Predicting Reaction types

S, D, SD, DD, C, or NR

Indicate the type of reaction (or NR for no reaction,) complete the word equation, and write a balanced chemical equation for each of the following. Make sure to include phase symbols. ~~\* include state of matter~~

SD 1. Aluminum metal and hydrochloric acid →



DD 2. Calcium hydroxide and nitric acid →



~~\* NR~~ 3. Aluminum and Magnesium →



S 4. Mercury(II) and oxygen →



DD 5. Zinc chloride and hydrogen sulfide →



NR 6. Silver chloride and sodium nitrate →



D 7. Sodium chlorate is heated →



DD 8. Barium nitrate and sodium chromate →



DD 9. Sodium bromide and silver nitrate →



DD 10. Calcium phosphate and aluminum sulfate →



D 11. Zinc carbonate is heated →



S 12. Potassium and fluorine →



S, C 13. Hydrogen and oxygen →



NR 14. Lead(II) and potassium chlorate →



S 15. Sodium and phosphoric acid →



# Writing Equations Predicting Reactions Answer Key

**Wolfgang Guggemos**



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