

Writing Complete Equations Practice

For each of the following problems, write complete chemical equations to describe the chemical process taking place. (Inorganic ions that are *not* organic processes or that affect no conditions, you don't write an equation for in chemical process)

- 1) Silver chloride (hydroxide pellets are added) to a solution of sulfuric acid. Silver sulfate and water are formed.
- 2) When dry water is heated for purification purposes, the temperature is brought up to 200° C for 25 minutes.
- 3) If a copper rod is placed into a solution of silver nitrate, silver crystals form on the surface of the copper. Additionally, highly soluble copper (II) nitrate is generated.
- 4) When compounds $\text{C}_2\text{H}_6\text{O}_2$ is burned in oxygen, carbon dioxide and water vapor are formed.
- 5) When a chunk of potassium metal is ground into a very fine powder and heated to drive off any atmospheric moisture, the resulting powder is an excellent catalyst for chemical reactions.

Writing Complete Equations Practice Answers For Teachers

Katrin Zwirgmaier



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