#### Types of Chemical Reaction and Predicting Products Worksheet

Balance the following. Indicate which type of chemical reaction (synthesis, decomposition, singledisplacement, double-displacement, acid-base, combustion, precipitation, or gas evolution) is being represented in 1 to 13.

- Na<sub>3</sub>PO<sub>4</sub> (aq)+ KOH(aq) → NaOH(aq) + K<sub>3</sub>PO<sub>4</sub>(s) Reaction Type
- MgCl<sub>2 (S)</sub> + \_\_\_Li<sub>2</sub>CO<sub>3(t)</sub> → \_\_\_MgCO<sub>3(t)</sub> + \_\_\_LiCl(aq) Reaction Type
- C<sub>6</sub>H<sub>12(1)</sub> + O<sub>2(q)</sub> → CO<sub>2(q)</sub> + H<sub>2</sub>O(I) Reaction Type
- Pb(s) + FeSO<sub>4(sq)</sub> → PbSO<sub>4(s)</sub> + Fe(s) Reaction Type \_\_\_\_\_
- CaCO<sub>3(s)</sub> → CaO(s) + CO<sub>2</sub> (g) Reaction Type
- P<sub>4(s)</sub> + O<sub>2(g)</sub> → P<sub>2</sub>O<sub>3(s)</sub> Reaction Type \_\_\_\_\_
- RbNO<sub>3(aq)</sub> + BeF<sub>2(aq)</sub> → Be(NO<sub>3</sub>)<sub>2(aq)</sub> + RbF(aq) Reaction Type \_\_\_\_\_
- AgNO<sub>3(aq)</sub> + Cu(s) → Cu(NO<sub>3</sub>)<sub>2(aq)</sub> + Ag(s) Reaction Type \_\_\_\_\_
- 9.  $C_3H_6O(I) + O_{2(g)} \rightarrow CO_{2(g)} + H_2O(I)$
- C<sub>5</sub>H<sub>5(0</sub> + Fe(S) → Fe(C<sub>5</sub>H<sub>5</sub>)<sub>2</sub> (S) Reaction Type \_\_\_\_\_
- SeCl<sub>6</sub> (s)+ O<sub>2(g)</sub> → SeO<sub>2(s)</sub> + Cl<sub>2(g)</sub>
- Reaction Type \_\_\_\_\_
- MgI<sub>2(aq)</sub> + Mn(SO<sub>3</sub>)<sub>2(s)</sub> → MgSO<sub>3(s)</sub> + MnI<sub>4</sub>(aq) Reaction Type
- NO<sub>2(g)</sub> → O<sub>2(g)</sub> + N<sub>2(g)</sub>

Reaction Type

Reaction Type \_\_\_\_\_

# **Ws 4 Balancing Equations And Predicting Products**

**M** Mosston

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U.S. Government Research Reports ,1964 Chemistry Homework Frank Schaffer Publications, Joan
DiStasio,1996-03 Includes the periodic table writing formulas balancing equations stoichiometry problems and more
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literature is presented Subjects covered under heat transfer and fluid flow include radioinduced heating boiling boiler evaporators pump and heat exchanger design hydrodynamics coolants and their properties thermal and flow instrumentation high temperature materials thermal properties of materials and thermal insulation Subjects covered less completely include thermodynamics aerodynamics high temperature corrosion corrosion specific to heat transfer systems erosion mass transfer corrosion film formation and effects coolant processing and radioactivity radiation effects of heat transfer materials and pertinent data of thermonuclear processes Subject report number availability and author indexes are given

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