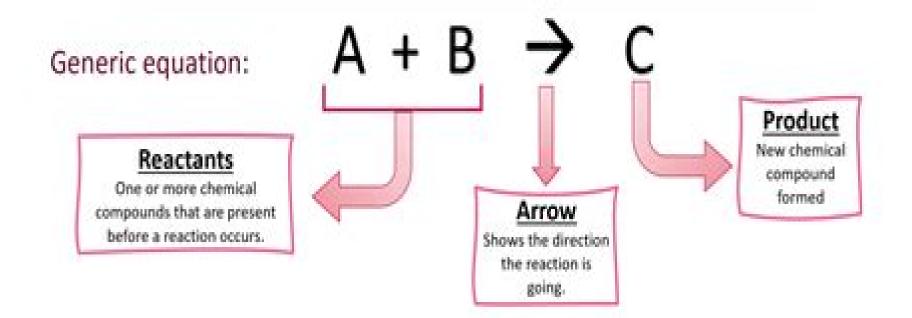
# **General Chemical Reaction Equations**

Chemical reactions are often represented by equations. These equations give information about the reaction in a simplified way.



$$\begin{array}{ccc} & \underline{\text{Vinegar}} & \underline{\text{Baking Soda}} & \underline{\text{Salt}} & \underline{\text{Gas}} & \underline{\text{Water}} \\ \text{Example: } \text{CH}_3\text{OOH} + \text{NaHCO}_3 & \xrightarrow{} \text{CH}_3\text{COONa} + \text{CO}_2 + \text{H}_2\text{O} \end{array}$$

An example reaction is vinegar/acetic acid reacting with baking soda/sodium bicarbonate to form a salt/sodium acetate, gas/carbon dioxide, and water. This reaction can safely be done at home using ingredients found in your kitchen!

# **Writing Formulas From Reaction Lab Conclusion**

J. A. Beran, Mark Lassiter

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