

Name: _____



Balancing Chemical Equations

Directions: Balance each of the following chemical equations.

- 1) ____ $C_{30}H_{42} + O_2 \rightarrow CO_2 + H_2O$
- 2) ____ $Al(OH)_3 + HBr \rightarrow AlBr_3 + H_2O$
- 3) ____ $CH_3CH_2CH_2CH_3 + O_2 \rightarrow CO_2 + H_2O$
- 4) ____ $C_3H_8 + O_2 \rightarrow CO_2 + H_2O$
- 5) ____ $Li + AlCl_3 \rightarrow LiCl + Al$
- 6) ____ $C_3H_8 + O_2 \rightarrow CO_2 + H_2O$
- 7) ____ $NH_4OH + H_3PO_4 \rightarrow (NH_4)_3PO_4 + H_2O$
- 8) ____ $Rb + P \rightarrow Rb_3P$
- 9) ____ $CH_4 + O_2 \rightarrow CO_2 + H_2O$
- 10) ____ $Al(OH)_3 + H_2SO_4 \rightarrow Al_2(SO_4)_3 + H_2O$
- 11) ____ $Na + Cl_2 \rightarrow NaCl$
- 12) ____ $Rb + S_8 \rightarrow Rb_2S$
- 13) ____ $H_3PO_4 + Ca(OH)_2 \rightarrow Ca_3(PO_4)_2 + H_2O$
- 14) ____ $NH_3 + HCl \rightarrow NH_4Cl$
- 15) ____ $Li + H_2O \rightarrow LiOH + H_2$
- 16) ____ $Ca_3(PO_4)_2 + SiO_2 + C \rightarrow CaSiO_3 + CO + P$
- 17) ____ $NH_3 + O_2 \rightarrow N_2 + H_2O$
- 18) ____ $FeS_2 + O_2 \rightarrow Fe_2O_3 + SO_2$

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