

BALANCING CHEMICAL EQUATIONS

Name Grove

Rewrite and balance the equations below.

- $\text{N}_2 + \text{H}_2 \rightarrow \text{NH}_3$ $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$
- $\text{KClO}_3 \rightarrow \text{KCl} + \text{O}_2$ $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$
- $\text{NaCl} + \text{F}_2 \rightarrow \text{NaF} + \text{Cl}_2$ $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$
- $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
- $\text{AgNO}_3 + \text{MgCl}_2 \rightarrow \text{AgCl} + \text{Mg}(\text{NO}_3)_2$ $2\text{AgNO}_3 + \text{MgCl}_2 \rightarrow 2\text{AgCl} + \text{Mg}(\text{NO}_3)_2$
- $\text{AlBr}_3 + \text{K}_2\text{SO}_4 \rightarrow \text{KBr} + \text{Al}_2(\text{SO}_4)_3$ $2\text{AlBr}_3 + 3\text{K}_2\text{SO}_4 \rightarrow 6\text{KBr} + \text{Al}_2(\text{SO}_4)_3$
- $\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$
- $\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ $\text{C}_3\text{H}_8 + 5\text{O}_2 \rightarrow 3\text{CO}_2 + 4\text{H}_2\text{O}$
- $\text{C}_8\text{H}_{18} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ $\text{C}_8\text{H}_{18} + \frac{25}{2}\text{O}_2 \rightarrow 8\text{CO}_2 + 9\text{H}_2\text{O}$
- $\text{FeCl}_3 + \text{NaOH} \rightarrow \text{Fe}(\text{OH})_3 + \text{NaCl}$ $\text{FeCl}_3 + 3\text{NaOH} \rightarrow \text{Fe}(\text{OH})_3 + 3\text{NaCl}$
- $\text{P} + \text{O}_2 \rightarrow \text{P}_2\text{O}_5$ $4\text{P} + 5\text{O}_2 \rightarrow 2\text{P}_2\text{O}_5$
- $\text{Na} + \text{H}_2\text{O} \rightarrow \text{NaOH} + \text{H}_2$ $2\text{Na} + 2\text{H}_2\text{O} \rightarrow 2\text{NaOH} + \text{H}_2$
- $\text{Ag}_2\text{O} \rightarrow \text{Ag} + \text{O}_2$ $2\text{Ag}_2\text{O} \rightarrow 4\text{Ag} + \text{O}_2$
- $\text{S}_8 + \text{O}_2 \rightarrow \text{SO}_3$ $\text{S}_8 + 12\text{O}_2 \rightarrow 8\text{SO}_3$
- $\text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + \text{O}_2$ $6\text{CO}_2 + 6\text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$
- $\text{K} + \text{MgBr}_2 \rightarrow \text{KBr} + \text{Mg}$ $2\text{K} + \text{MgBr}_2 \rightarrow 2\text{KBr} + \text{Mg}$
- $\text{HCl} + \text{CaCO}_3 \rightarrow \text{CaCl}_2 + \text{H}_2\text{O} + \text{CO}_2$ $2\text{HCl} + \text{CaCO}_3 \rightarrow \text{CaCl}_2 + \text{H}_2\text{O} + \text{CO}_2$

Writing And Balancing Chemical Equations Answers

Morris Hein, Susan Arena



Writing And Balancing Chemical Equations Answers:

National 5 Chemistry with Answers, Second Edition Barry McBride, Stephen Jeffrey, John Anderson, Paul McCranor, Fran Macdonald, 2018-07-30 Exam Board SQA Level National 5 Subject Chemistry First Teaching September 2017 First Exam Summer 2018 The second edition of this textbook has been fully revised and updated to reflect changes made to the SQA syllabus from 2017 onwards New features include Refreshed content Additional candidate advice Model answers for open ended questions

Foundations of College Chemistry, Alternate Morris Hein, Susan Arena, 2010-01-26 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years this book has helped them master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis

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eBook: General, Organic and Biological Chemistry 2e SMITH, 2012-02-16 eBook General Organic and Biological Chemistry 2e

Quantitative Chemical Analysis, Sixth Edition Daniel C. Harris, 2003 For instructors who wish to focus on practical industrial or research chemistry Includes case studies applications boxes and spreadsheet applications

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Morris Hein,Susan Arena,Cary Willard,2016-08-02 This text is an unbound three hole punched version Used by over 750 000 students Foundations of College Chemistry Binder Ready Version 15th Edition is praised for its accuracy clear no nonsense approach and direct writing style Foundations direct and straightforward explanations focus on problem solving making it the most dependable text on the market Its comprehensive scope proven track record outstanding in text examples and problem sets were all designed to provide instructors with a solid text while not overwhelming students in a difficult course Foundations fits into the prep intro chemistry courses which often include a wide mix of students from science majors not yet ready for general chemistry allied health students in their 1st semester of a GOB sequence science education students for elementary school teachers to the occasional liberal arts student fulfilling a science requirement Foundations was specifically designed to meet this wide array of needs

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