

FORMULAS FOR IONIC COMPOUNDS

1. Write the chemical formula for the ionic compounds made from the following elements.

Elements	metal ion	non-metal ion	Chemical Formula	Name
potassium and iodine	K^{+1}	I^{-1}	KI	Potassium Iodide
barium and chlorine	Ba^{2+}	Cl^{-1}	$BaCl_2$	Barium Chloride
sodium and oxygen	Na^{+1}	O^{2-}	Na_2O	sodium Oxide
aluminum and nitrogen	Al^{3+}	N^{3-}	AlN	Aluminum Nitride
lithium and sulfur	Li^{+1}	S^{2-}	Li_2S	Lithium Sulfide
magnesium and oxygen	Mg^{+2}	O^{2-}	MgO	Magnesium Oxide
calcium and chlorine	Ca^{2+}	Cl^{-1}	$CaCl_2$	calcium chloride
boron and chlorine	B^{3+}	Cl^{-1}	BCl_3	boron chloride
beryllium and sulfur	Be^{2+}	S^{2-}	BeS	beryllium sulfide

2. Complete the chart below and write the ionic formulas to complete it. A few have been done for you.

	F	Cl	O ²⁻	S ²⁻	N ³⁻	P ³⁻
Li ⁺	LiF	LiCl	Li ₂ O	Li ₂ S	Li ₃ N	Li ₃ P
Na ⁺	NaF	NaCl	Na ₂ O	Na ₂ S	Na ₃ N	Na ₃ P
K ⁺	KF	KCl	K ₂ O	K ₂ S	K ₃ N	K ₃ P
Be ²⁺	BeF ₂	BeCl ₂	BeO	BeS	Be ₃ N ₂	Be ₃ P ₂
Mg ²⁺	MgF ₂	MgCl ₂	MgO	MgS	Mg ₃ N ₂	Mg ₃ P ₂
Ca ²⁺	CaF ₂	CaCl ₂	CaO	CaS	Ca ₃ N ₂	Ca ₃ P ₂
Al ³⁺	AlF ₃	AlCl ₃	Al ₂ O ₃	Al ₂ S ₃	AlN	AlP
Ga ³⁺	GaF ₃	GaCl ₃	Ga ₂ O ₃	Ga ₂ S ₃	GaN	GaP
B ³⁺	BF ₃	BCl ₃	B ₂ O ₃	B ₂ S ₃	BN	BP

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Table of Contents Writing Ionic Formulas Lab 207 Answer Key

1. Understanding the eBook Writing Ionic Formulas Lab 207 Answer Key
 - The Rise of Digital Reading Writing Ionic Formulas Lab 207 Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Writing Ionic Formulas Lab 207 Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Writing Ionic Formulas Lab 207 Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Writing Ionic Formulas Lab 207 Answer Key
 - Personalized Recommendations

- Writing Ionic Formulas Lab 207 Answer Key User Reviews and Ratings
- Writing Ionic Formulas Lab 207 Answer Key and Bestseller Lists
- 5. Accessing Writing Ionic Formulas Lab 207 Answer Key Free and Paid eBooks
 - Writing Ionic Formulas Lab 207 Answer Key Public Domain eBooks
 - Writing Ionic Formulas Lab 207 Answer Key eBook Subscription Services
 - Writing Ionic Formulas Lab 207 Answer Key Budget-Friendly Options
- 6. Navigating Writing Ionic Formulas Lab 207 Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Writing Ionic Formulas Lab 207 Answer Key Compatibility with Devices
 - Writing Ionic Formulas Lab 207 Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Writing Ionic Formulas Lab 207 Answer Key
 - Highlighting and Note-Taking Writing Ionic Formulas Lab 207 Answer Key
 - Interactive Elements Writing Ionic Formulas Lab 207 Answer Key
- 8. Staying Engaged with Writing Ionic Formulas Lab 207 Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Writing Ionic Formulas Lab 207 Answer Key
- 9. Balancing eBooks and Physical Books Writing Ionic Formulas Lab 207 Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Writing Ionic Formulas Lab 207 Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Writing Ionic Formulas Lab 207 Answer Key
 - Setting Reading Goals Writing Ionic Formulas Lab 207 Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Writing Ionic Formulas Lab 207 Answer Key
 - Fact-Checking eBook Content of Writing Ionic Formulas Lab 207 Answer Key

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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