

## Writing Complete Equations Practice

For each of the following problems, write complete chemical equations to describe the chemical process taking place. (Inorganic ions that are *not* organic processes or that affect no conditions, you don't write an equation for in chemical process)

- 1) Silver chloride (hydroxide pellets are added) to a solution of sulfuric acid. Silver sulfate and water are formed.
- 2) When dry water is heated for purification purposes, the temperature is brought up to 200° C for 25 minutes.
- 3) If a copper rod is placed into a solution of silver nitrate, silver crystals form on the surface of the copper. Additionally, highly soluble copper (II) nitrate is generated.
- 4) When compounds  $\text{C}_2\text{H}_6\text{O}_2$  is burned in oxygen, carbon dioxide and water vapor are formed.
- 5) When a chunk of potassium metal is ground into a very fine powder and heated to drive off any atmospheric moisture, the resulting powder is an extremely resistant for chemical reactions.

# Writing Complete Equations Practice Chemfiesta

**Marcia L. Gillette, H. Anthony Neidig**



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trois états de la matière bien connue sous la forme de glace ou de neige l'eau solide est également présente dans d'innombrables minéraux les hydrates naturels

la nature de l'eau - Jan 28 2022

web l'eau est une passerelle entre les hommes elle se singularise par sa nature transversale il existe ainsi différents mondes de l'eau qui ouvrent aux multiples regards que nous portons sur elle scientifique artistique économique