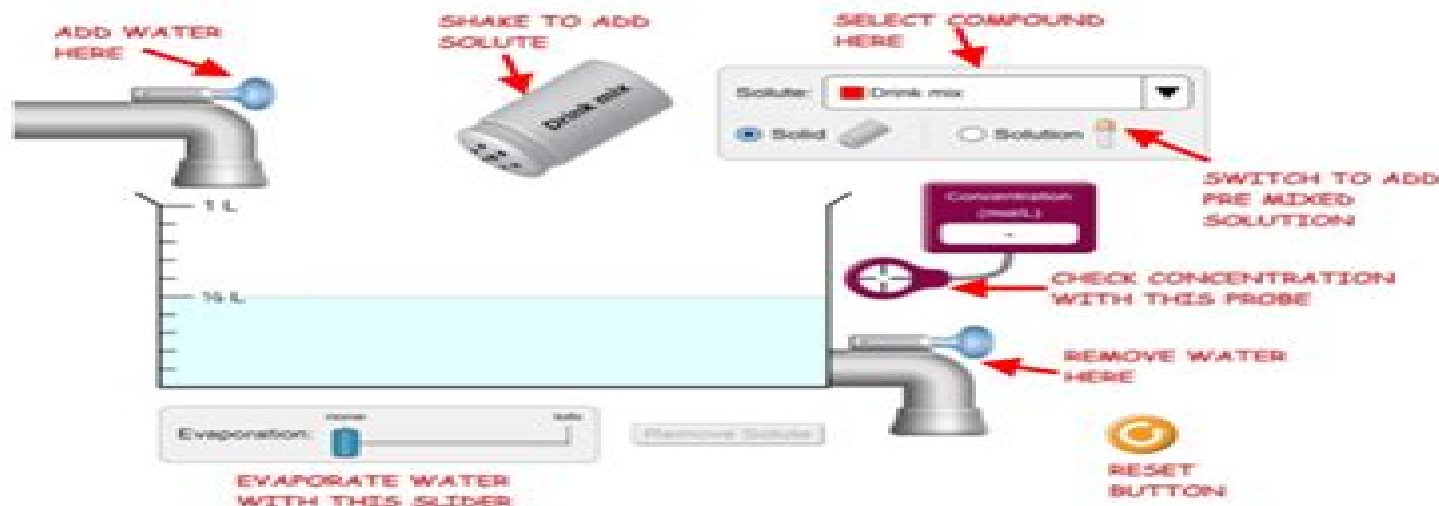


Molarity Virtual Lab

https://phet.colorado.edu/sims/html/concentration/latest/concentration_en.html

Familiarize yourself with the operation of the various functions of the simulation. You can always press the RESET button to get back to the original situation.



Press Reset. Move the Concentration Probe into the liquid. Then Shake some Drink mix into the water. Continue to add More Solute.

PART 1 – Effect of Changing Amount of Solute/Solvent

1. What happens to the concentration as more solute is added? [Increase]
2. What is the unit on the meter measuring Concentration? [mol/L]
3. What happens to the color as more solute is added? [Turns red]
4. What is the relationship between concentration and color? [Higher concentration of red the solution will become.]
5. Now add water. What happens to the concentration as the quantity of water increases? [lower] Explain why. (Mention what is happening to the quantity of solvent and solute) [The concentration decreases as water is added because the solvent increases while the amount of solute remains the same resulting in a lower concentration.]
6. Now Remove water with the valve on the lower right. What happens to the concentration as the quantity of water decreases? [stays the same] Explain why. (Mention what is happening to the quantity of solvent and solute) [The concentration stays the same as the quantity of water decreases as the amount of solute remains unchanged while the solvent decreases.]

Virtual Lab Answer Key

Antonio J. Tallón-Ballesteros

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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