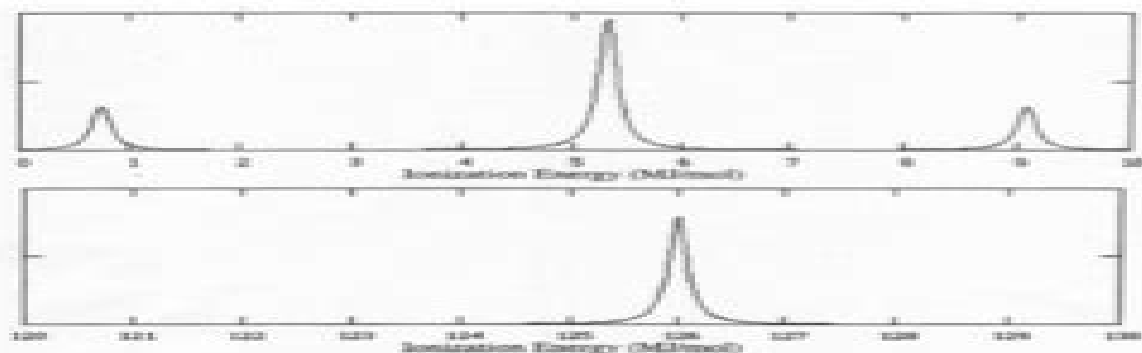


Data C. Simulated Photoelectron Spectrum of Unknown #1

Photon Energy: 1460 MJ/mol



Analysis and Calculations C

Note: The two portions of the PES spectrum are drawn to different scales on both axes. Only the portions of the spectrum containing peaks are shown.

Ionization Energy (MJ/mol)	Area of Peak
0.74	3.5
5.31	104
9.07	3.5
126.0	2.8

1. In the spectrum above assign sublevels to each peak.
2. Determine the *relative* number of electrons in each of the sublevels.
3. Using your knowledge of electron configurations determine the actual number of electrons in each sublevel. (Hint: Your answer for #2 and #3 should be different. If they are not it is: you should ask yourself the difference between "relative" number and "actual" number.)
4. Identify the element used in this PES spectrum.

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