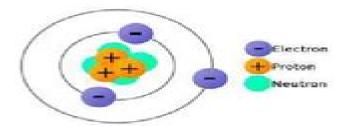


#### **Atomic Structure Review**

- 1. Mass number = \_\_\_\_\_
- 2. Give the charge of each: Protons = Neutrons = Electrons =
- The number of protons is the same as
- If atoms have the same number of valence electrons then they will have similar......
- 5. The nucleus has a \_\_\_\_\_ charge and contains \_\_\_\_\_
- 6. Electrons are located \_\_\_\_\_ of the \_\_\_\_\_ and are found in the \_\_\_\_\_
- 7. Order from lightest to heaviest:

Electrons, Protons, Nucleus, Neutrons

8.



Protons = \_\_\_\_\_

Neutrons = \_\_\_\_\_

Electrons = \_\_\_\_\_

Atomic number = \_\_\_\_\_

Mass number = \_\_\_\_\_

Valence e = \_\_\_\_\_

Charge = \_\_\_\_\_

9. 20

Ne

10 Protons = \_\_\_\_ Neutrons = \_\_\_\_ Electrons = \_\_\_\_ Mass # = \_\_

Atomic # = \_\_\_\_\_ Valence e- = \_\_\_\_ Charge = \_\_\_\_\_

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