

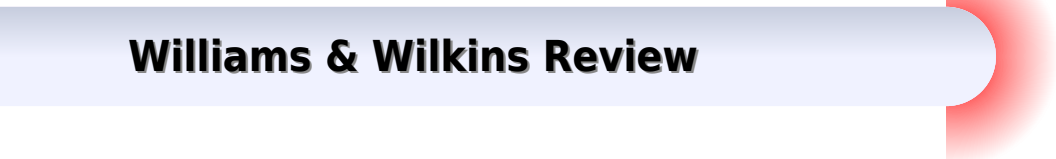
assume a neutral atom

You can become more familiar with the atomic structure of some common substances by completing the chart below. For each substance, you have been given enough information to fill in all the blanks.

Substance	Symbol	Atomic Number	(p+n) Mass Number	Number of Protons	Number of Neutrons	Number of Electrons
Helium	He	2	4	2	2	2
Magnesium	Mg	12	24	12	12	12
Zinc	Zn	30	65	30	35	30
Bromine	Br	35	80	35	45	35
Aluminum	Al	13	27	13	14	13
Uranium	U	92	238	92	146	92
Sodium	Na	11	23	11	12	11
Krypton	Kr	36	84	36	48	36
Calcium	Ca	20	40	20	20	20
Silver	Ag	47	108	47	61	47

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representation the macro the empirical properties of substances the sub micro the natures of the entities giving rise to those properties and the symbolic the number of entities involved in any changes that take place Although understanding this triplet relationship is a key aspect of chemical education there is considerable evidence that students find great difficulty in achieving mastery of the ideas involved In bringing together the work of leading chemistry educators who are researching the triplet relationship at the secondary and university levels the book discusses the learning involved the problems that students encounter and successful approaches to teaching Based on the reported research the editors argue for a coherent model for understanding the triplet relationship in chemical education *The Software Encyclopedia* ,1988 **College**

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