WORKSHEET ON SOLUTION CONCENTRATIONS

INTRODUCTION

A solution is made up of two parts, the **SOLVENT** and the **SOLUTE**. The solution could be **CONCENTRATED** or **DILUTE**.

SOLVENT	The liquid which dissolves the solute. It is usually water.		
SOLUTE	The substance which dissolves - often a solid, but not always.		
CONCENTRATED	There is a lot of solute and a little water.		
DILUTE	There is a lot of water and only a little solute.		

When we measure out 50cm³ of a solution, e.g., hydrochloric acid, we measure out some solvent (water) and some acid together. In any reactions of the acid, it is the HCl particles which are involved. The water is just a carrier for the acid, so when we measure out a volume of the solution, we want to know how much acid it contains. To do this, we need to define the CONCENTRATION of the solution. This tells us how much solute there is in a fixed volume of the solution. In chemistry, "how much" means MOLES and the "fixed volume" is taken as 1 cubic decimetre (1dm³).

[NB: Idm³ is 1000cm³ and is the same volume as 1 litre.]

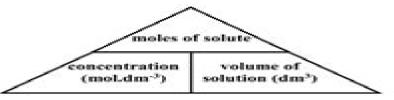
DEFINITION

Concentration of solution = (moles per dm³)		number of moles of solute
		volume of solution in dm ³

- You must learn this relationship and be able to use it and rearrange it to calculate moles
 of solute or volume of solution, given the other two quantities.
- Remember that the volume must be in dm³, so convert from cm³ if necessary.
- The units for concentration are moles per cubic decimetre which is written mol.dm⁻¹
- Lab bottles use the letter M (meaning molar) and this is the same thing as mol.dm³

RELATIONSHIPS

Use the triangle opposite to help you answer the questions which follow. Some worked examples are given to help you.



Examples:-

- I) What is the concentration of a solution which contains 2 moles of copper (II) sulphate in 4dm³ of solution?
 - Ans: concentration = moles + volume = $2 + 4 = 0.5 \text{ mol.dm}^3$
- 2) How many moles of sodium hydroxide are there in 200cm³ of a solution of concentration 0.2 mol.dm⁻³? (Remember that 200cm³ = 200 + 1000 = 0.2dm³)
 Ans: moles = concentration × volume = 0.2 × 0.2 = 0.04 moles

Skill Practice 39 Concentration Practice Answers

Sabine Zange

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Exercise Physiology William J. Kraemer, Steven J. Fleck, Michael R. Deschenes, 2011-03-01 Designed for undergraduate course work this exercise physiology textbook unites research and theory with real world application so students can easily relate to the concepts being presented The unique applied approach fully engages you in discovering how the human body works and responds to exercise You ll not only gain a solid foundation in exercise physiology concepts you ll also learn how to apply these concepts on the job to optimize athletic performance and well being Moreover you ll come to understand the vital health benefits of exercise and physical activity for all individuals at all ages including special populations Beginning with basic exercise physiology concepts the text progressively builds your knowledge by integrating these concepts into practical discussions of nutrition and training The text stresses a research based approach enabling you to locate and evaluate the evidence you need to make good decisions Numerous examples further underscore the importance of basic concepts and research in addressing real life challenges in exercise and athletic training Math Practice Simplified: Word Problems (Book I) Ann Cassill Sofge, M.A., 2021-06-04 Strong math skills are essential to success in school and life Math Practice Simplified Word Problems provides practice in solving word problems from those that involve simple addition to those with more complex computations with money and percentages Basic concepts involving whole numbers fractions money and percentages make up the majority of the word problems in this eBook Time graphs and maps are also dealt with There are problems with both metric and customary units of measure The problems become progressively more challenging within each section The sections themselves however are not ordered according to difficulty Answers are provided at the back of the Preparing Students for Testing and Doing Better in School Rona F. Flippo, 2008-01-31 Improve students test taking book success by teaching effective time management strategies organizational skills note taking techniques ways to handle test anxiety and methods for taking essay and objective tests Math Practice Simplified I Ann Cassill Sofge, 2024-12-31 This resource is for grades 6 8 and helps all students build the mental facility needed to do well on standardized tests and to move on to higher level thinking Strong math skills are essential to success in school and life Math Practice Simplified Word Problems provides practice in solving word problems from those that involve simple addition to those with more complex computations with money and percentages Basic concepts involving whole numbers fractions money and percentages make up the majority of the word problems in this resource Time graphs and maps are also dealt with There are problems with both metric and customary units of measure The problems become progressively more challenging within each section The sections themselves however are not ordered according to difficulty Answers are provided at the back of the book

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Scholarly Writings Dr. Susan Ramilo-Butac, 2023-09-08 BITES A Compilation of Scholarly Writings by Dr Susan Ramilo Butac

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