#### THE PERIODIC TABLE CHAPTER 5 TEST: MULTIPLE CHOICE Identify the choice that best completes the statement or answers the question. The order of elements in the periodic table is based on the number of protons in the nucleus. the electric charge of the nucleus. the number of neutrons in the nucleus. atomic mass. 2. Atoms of elements that are in the same group have the same number of protons. e. valence electrons. 200 d. protons and neutrons. neutrons. 31. Which of the following elements is an alkali metal? calleium. c. mercury magnesium d. sodirum Semiconductors are elements that have large atomic masses but small atomic numbers. b. do not form compounds. c. can conduct heat and electricity under certain conditions. are extremely hard. 5. Carbon and other nonmetals are found in which area of the periodic table? on the left-most side on the right side in the middle column of the periodic table in the bottom rows. In Mendeleev's periodic table, elements in each column had similar atomic masses. e. atomic numbers properties. d. symbolis. 7. Magnesium (Mg) is located to the right of sodium (Na) because Mg has fewer protons. c. no protons. no neutrons. d. more protons. As you move from left to right across the periodic table, elements become less metallic. c. have a lower atomic weight. b. have a lower atomic number. d. become more metallic. How was Mendeleev's periodic table arranged?

e. by increasing atomic number

d. by decreasing atomic number

a. by increasing atomic mass

b. by decreasing atomic mass

mainne,

HPS

date:

# **Test B Chem Holt Mcdougal Modern Chem**

**G** Orfield

Test B Chem	<b>Holt Mcdougal</b>	<b>Modern Chem:</b>
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