

Chapter 6 Practice Test: Chemical Bonding

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. The charge on an ion is
a. always positive.
b. always negative.
c. either positive or negative.
d. zero.
- _____ 2. According to the octet rule, a calcium atom has a tendency to
a. lose one electron.
b. lose two electrons.
c. gain one electron.
d. gain two electrons.
- _____ 3. If a compound forms by ionic bonding, which is *not* true?
a. A positively charged atom or group of atoms attracts a negatively charged atom or group of atoms.
b. The net charge of the compound is zero.
c. The compound contains just two atoms, each of opposite charge.
d. Several ions group together in a tightly packed structure.
- _____ 4. The only property listed that is *not* characteristic of ionic compounds is
a. high melting point.
b. hardness.
c. lack of crystal structure.
d. good conductor of electricity.
- _____ 5. Which formula listed below represents a polyatomic ion?
a. NH_4^+
b. H_2SO_4
c. NaCl
d. H_2O
- _____ 6. The melting points of ionic compounds are higher than the melting points of molecular compounds because
a. ionic substances tend to vaporize at room temperature.
b. ionic substances are brittle.
c. attractive forces between ions are greater than the attractive forces between molecules.
d. the numbers of positive and negative charges are equal in an ionic compound.

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Ying Liu



Test B Ch 6 Chemical Bonds Holt:

Whispering the Strategies of Language: An Psychological Journey through **Test B Ch 6 Chemical Bonds Holt**

In a digitally-driven earth wherever displays reign supreme and instant conversation drowns out the subtleties of language, the profound strategies and psychological nuances hidden within words usually get unheard. Yet, located within the pages of **Test B Ch 6 Chemical Bonds Holt** a charming literary value pulsing with fresh thoughts, lies an exceptional quest waiting to be undertaken. Penned by a skilled wordsmith, this wonderful opus invites readers on an introspective journey, lightly unraveling the veiled truths and profound impact resonating within the very material of every word. Within the psychological depths with this touching review, we will embark upon a heartfelt exploration of the book is primary themes, dissect their fascinating publishing style, and succumb to the effective resonance it evokes serious within the recesses of readers hearts.

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