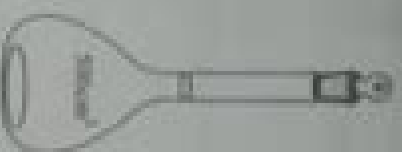


- 8.3 A learner prepares a standard solution by dissolving 5.3g of sodium carbonate, Na_2CO_3 in 500cm^3 of distilled water using the apparatus shown below.



- 8.3.1 Write down the name of the apparatus shown above.

(1)

- 8.3.2 Calculate the concentration of the sodium carbonate solution.

(4)

- 8.4 The learner used standard solution prepared in 8.3 to determine the concentration of a sample of hydrochloric acid (HCl). 25cm^3 of the sodium carbonate solution and added a few drops of phenolphthalein indicator. $18,65\text{cm}^3$ of the Hydrochloric acid reacted completely with the sodium carbonate. The reaction that takes place is:



- 8.4.1 What is the colour change at endpoint?

(2)

- 8.4.2 Calculate the concentration of the hydrochloric acid sample.

(3)

[24]

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