Six Types of Chemical Reaction Worksheet

Answers

Balance the following reactions and indicate which of the six types of chemical reaction are being represented:

- NaBr + 1 Ca(OH)₂ → 1 CaBr₂ + 2 NaOH
 Type of reaction: double displacement
- 2) 2 NH₃+ 1 H₂SO₄ → 1 (NH₄)₂SO₄ Type of reaction: synthesis
- 3) 4 C₅H₉O + 27 O₂ → 20 CO₂ + 18 H₂O
 Type of reaction: combustion
- 4) 3 Pb + 2 H₃PO₄ → 3 H₂ + 1 Pb₃(PO₄)₂
 Type of reaction: single displacement
- 5) 1 Li₃N + 3 NH₄NO₃ → 3 LiNO₃ + 1 (NH₄)₃N Type of reaction: double displacement
- 6) 3 HBr + 1 Al(OH)₃ → 3 H₂O + 1 AlBr₃

 Type of reaction: acid-base
- 7) What's the main difference between a double displacement reaction and an acid-base reaction? Acid-base reactions form water.
- 8) Combustion reactions always result in the formation of water. What other types of chemical reaction may result in the formation of water? Write examples of these reactions on the opposite side of this paper. Acid-base: HCl + NaOH → H₂O + NaCl Synthesis: 2 H₂ + O₂ → 2 H₂O Decomposition: NH₄OH → NH₃ + H₂O Single displacement: H₂ + 2 NaOH → 2 Na + 2 H₂O (not common)

Six Types Of Reactions Homework Answers

Dr M M Sharma

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